SYNTHESIS OF ASPIRIN

I. OBJECTIVES AND BACKGROUND

You will:

- synthesize acetylsalicylic acid (aspirin) by carrying out a simple organic reaction,
- separate your product from the reaction mixture by vacuum filtration,
- · purify your product by recrystallization.
- perform a chemical test to identify the change in functional group from reactant to product, and
- determine the success of your synthesis by calculating the percentage yield of your product.

INTRODUCTION

Aspirin is one of the most widely used medications in the world. It is employed as an analgesic (pain relief), an anti-pyretic (fever control) and an anti-inflammatory. More recently, studies have indicated that daily intake of small doses of aspirin can lower the risk of heart attack and stroke in high-risk patients.

The history of aspirin and its precursor dates back to ancient times. Documents attributed to Hippocrates, the father of modern medicine, from the 4th century B.C. refer to the alleviation of pain by chewing on the bark of a willow tree or ingesting a powder made from the bark and leaves of the willow. This remedy was passed on from generation to generation.

Fast forward now to the 19th century, where the field of organic chemistry began to experience tremendous growth. By 1838, chemists had managed to isolate, purify and identify the component of willow bark that provided the analgesic benefit. The compound was named salicylic acid, which was based on the genus name of the willow.

Aspirin Synthesis Lab Manuals

ML Morrison

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