



Name: .....

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### Making Salts Worksheet

This worksheet accompanies *Making Salts Summary* and *Precipitation Reactions Summary*.

1. When an acid reacts with a base, a neutralisation reaction occurs, producing a salt. Using the following equation, name the salts that are produced in the acid-base reactions.

acid + base  $\longrightarrow$  salt

- a) hydrochloric acid + ammonia  $\longrightarrow$  .....
- b) nitric acid + sodium hydroxide  $\longrightarrow$  .....
- c) phosphoric acid + potassium hydroxide  $\longrightarrow$  .....



2. There are three other ways of making salts from acids. Complete the following equations to illustrate them.

- a) acid + metal  $\longrightarrow$  salt + .....
- b) acid + .....  $\longrightarrow$  salt + water
- c) acid + carbonate  $\longrightarrow$  salt + ..... + water

3. Use the above information to complete the following reactions that form soluble salts using acids.

- a) iron oxide + sulfuric acid  $\longrightarrow$  ..... + water
- b) ..... + hydrochloric acid  $\longrightarrow$  magnesium chloride +  $\text{H}_2\text{O}$  +  $\text{CO}_2$
- c) zinc + .....  $\longrightarrow$  zinc nitrate + hydrogen

4. Name two ways in which salt is used by humans in everyday life.

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.....

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# Acid Base Or Salt Physical Science If8767 Answers

**Padhraic Smyth**



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