

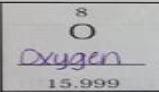
- Atomic Number
Symbol
Name
Atomic Mass

Atomic number equals
the number of

Protons or Electrons

Atomic mass equals

the number of protons + neutrons

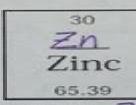


Atomic # = 8

Atomic Mass = 16

of Protons = 8

of Neutrons = 8



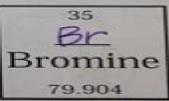
Atomic # = 30
Atomic Mass = 105
of Protons = 30
of Neutrons = 35
of Electrons = 30

Atomic # = 5
Atomic Mass = 7
of Protons = 3
of Neutrons = 4
of Electrons = 3

Silicon 28.086

Atomic # = 14
Atomic Mass = 28
of Protons = 14
of Neutrons = 14
of Electrons = 14

Atomic # = 5
Atomic Mass = 11
of Protons = 5
of Neutrons = 5
of Electrons = 5



Atomic # = 35

Atomic Mass = 80

of Protons = 35

of Neutrons = 45

of Electrons = 35

Atoms Family Atomic Math Challenge Answers

Richard Bailey

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