

Atomic Theory & Atom Structure: Multiple Choice

1. Which one of the following is not one of the postulates of Dalton's atomic theory?

- A) Atoms are composed of protons, neutrons, and electrons.
- B) All atoms of a given element are identical; the atoms of different elements are different and have different properties.
- C) Atoms of an element are not changed into different types of atoms by chemical reactions.
- D) atoms are neither created nor destroyed in chemical reactions.
- E) Compounds are formed when atoms of more than one element combine; a given compound always has the same relative number and kind of atoms.
- F) Each element is composed of extremely small particles called atoms.

ANS: A

1. Consider the following selected postulates of Dalton's atomic theory:

- (i) Each element is composed of extremely small particles called atoms.
- (ii) Atoms are indivisible.
- (iii) Atoms of a given element are identical.
- (iv) Atoms of different elements are different and have different properties.

1. Which of the postulates is/are no longer considered valid?

- A) (i) and (ii)
- B) (iii) only
- C) (ii) and (iii)
- D) (iii) only
- E) (iii) and (iv)

ANS: C

1. All atoms of a given element have the same _____.

- A) mass
- B) number of protons
- C) number of neutrons
- D) number of electrons and neutrons
- E) density

ANS: B

1. The nucleus of an atom does not contain _____.

- A) protons
- B) protons or neutrons
- C) neutrons
- D) subatomic particles
- E) electrons

ANS: E

Atomic Structure Multiple Choice Answers

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