

Webquest: Atomic Theories and Models

Answer these questions on your own, USING COMPLETE SENTENCES where appropriate (most of the questions, except tables and drawings). If you require more space, please complete on another piece of paper.

Atom Basics: Go to: <http://www.chemtutor.com/struct.html> and read the "And you thought you were strange" section to answer the following questions (put answers in the table).

1. What are the three subatomic particles that all atoms are made of?
2. Where are each of the three particles located within the atom?
3. What is the electrical charge of each particle?

1. The 3 subatomic particles	2. Location within the Atom	3. Electrical Charge
Electrons	Shells/Orbital's around Nucleus	Negative
Protons	Nucleus	Positive
Neutrons	Nucleus	Neutral

Early Ideas About Atoms: Go to <http://galileo.phys.virginia.edu/classes/252/atoms.html> and read the section on "Early Greek Ideas" in order to answer the following questions:

4. What was the "basic idea" about matter that Leucippus and Democritus proposed?

The basic idea was that if you could look at matter on smaller and smaller scales (which they of course couldn't) ultimately you would see individual atoms - objects that could not be divided further (that was the definition of atom).

5. How did they use atoms to explain different physical properties?

The different physical properties -- color, taste, and so on -- of materials came about because atoms in them had different shapes and/or arrangements and orientations with respect to each other.

6. How were the ideas of these two men received by Aristotle, and what was the result on the progress of atomic theory for the next 2,000 years?

350 B.C - Aristotle modified an earlier theory that matter was made of four "elements": earth, fire, water, air. This gave rise to the alchemists (who searched for a way to turn lead to gold).

John Dalton's Atomic Theory: Go to

<http://www.iun.edu/~cpanhd/C101webnotes/composition/dalton.html> and use the information there to answer the following questions:

7. When did Dalton form his Atomic Theory?

Dalton publish is Atomic Theory in 1808.

8. What are the four components of Dalton's Atomic Theory?

1) All matter is made of atoms. Atoms are indivisible and indestructible. 2) All atoms of a given element are identical in mass and properties. 3) Compounds are formed by a combination of two or more different kinds of atoms. 4) A chemical reaction is a rearrangement of atoms.

Atomic Webquest Answer Key

Rachel Sandford

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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