12

STOICHIOMETRY

Practice Problems

In your notations, pales the following problems.

SECTION 12.1 THE ARITHMETIC OF EQUATIONS

Ohe the 3-copy problem, soluting approach you featured to Obspary 1.

- An apple pin needs 10 large apples, 2 strates (top and bottom), and 1 sabineparen of closustace. Write a balanced regastion that fits this situation. How many apples are needed to make 26 piles?
- Two modes of points states ablieded and three modes of stagged are produced from the decomposition of two males of patentiam obligates, 8020(5):1. Write the finitesized equation. How many makes of oxygen are produced from 12 modes of points at the control.
- Using the equation from problem 2, how many makes of oxygen are produced from 14 reales of potassium chlomatic
- Two modecules of hydrogen reset with one molecule of oxyges to produce two molecules of water. How many molecules of water are produced from 2.8 × 10²⁴ molecules of oxygen? How many males of water are produced from 22.5 nodes of oxygen?

SECTION 12.2 CHEMICAL CALCULATIONS

 Calculate the number of males of hydrogen chloride produced from 10 moles of featnesses.

 Calculate the number of states of objective meeted to form 14 moles of insu(33) objectes.

 Culculate the number of grams of strongers disolds that are produced from a moles of sitric oxide.

$$2590 \log + O_{\rm phy} \rightarrow 2590 \rm phy$$

 Calculate the mass of orgges produced from the decomposition of PLO g of posterium chlorum.

- B. Calculate the mass of eliver needed to must with objection in produce 64 g of silver objected. (What Write a leakeneed equation first):
- How entity litters of authors momentale at STP are specied to return with 4.00 g; of oxygen gas to produce authors cliosiche?

 Calculate the number of lines of organ gas needed to produce 15.0 liters of distingen trioxide. Assume all gases are at the same conditions of temperature and pressure.

$$3N_{\rm e}kp = 30_{\rm e}kp \rightarrow 2N_{\rm e}0_{\rm e}kp$$

<u>Arithmetic Equations Of Stoichiometry Study Workbook</u> <u>Answers</u>

James E. Brady, Fred Senese

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Fredeen, Pearson, 2003-07 Multiple Solution Methods for Teaching Science in the Classroom Stephen DeMeo, 2008 For the first time in science education the subject of multiple solution methods is explored in book form While a multiple method teaching approach is utilized extensively in math education there are very few journal articles and no texts written on this topic in science Teaching multiple methods to science students in order to solve quantitative word problems is important for two reasons First it challenges the practice by teachers that one specific method should be used when solving problems Secondly it calls into question the belief that multiple methods would confuse students and retard their learning Using a case study approach and informed by research conducted by the author this book claims that providing students with a choice of methods as well as requiring additional methods as a way to validate results can be beneficial to student learning A close reading of the literature reveals that time spent on elucidating concepts rather than on algorithmic methodologies is a critical issue when trying to have students solve problems with understanding It is argued that conceptual understanding can be enhanced through the use of multiple methods in an environment where students can compare evaluate and verbally discuss competing methodologies through the facilitation of the instructor This book focuses on two very useful methods proportional reasoning PR and dimensional analysis DA These two methods are important because they can be used to solve a large number of problems in all of the four academic sciences biology chemistry physics and earth science This book concludes with a plan to integrate DA and PR into the academic science curriculum starting in late elementary school through to the introductory college level A challenge is presented to teachers as well as to textbook writers who rely on the single method paradigm to consider an alternative way to teach scientific problem solving Numerical Chemistry,

Problems and Problem Solving in Chemistry Education Georgios Tsaparlis, 2021

Problem solving is central to the teaching and learning of chemistry at secondary tertiary and post tertiary levels of education opening to students and professional chemists alike a whole new world for analysing data looking for patterns and making deductions As an important higher order thinking skill problem solving also constitutes a major research field in science education Relevant education research is an ongoing process with recent developments occurring not only in the area of quantitative computational problems but also in qualitative problem solving The following situations are considered some general others with a focus on specific areas of chemistry quantitative problems qualitative reasoning metacognition and resource activation deconstructing the problem solving process an overview of the working memory hypothesis reasoning with the electron pushing formalism scaffolding organic synthesis skills spectroscopy for structural characterization in organic chemistry enzyme kinetics problem solving in the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner

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