

ASSIGNING OXIDATION NUMBERS – WORKSHEET 1

Part A:

In the following questions, give the oxidation number of the indicated atoms/ion.

1. N in N_2O_3 _____

2. S in H_2SO_4 _____

3. C _____

4. C in CO _____

5. Na in NaCl _____

6. H in H_2O _____

7. Ba in BaCl_2 _____

8. N in NO_2^- _____

9. S in Al_2S_3 _____

10. S in HSO_4^- _____

11. Cl in $\text{Fe}(\text{ClO}_2)_3$ _____

12. Fe in $\text{Fe}(\text{ClO}_2)_3$ _____

13. N in NO_3^- _____

14. Cu^{2+} _____

15. Zn^{2+} _____

16. C in CH_4 _____

17. Mn in MnO_2 _____

18. S in SO_3^{2-} _____

19. Mg^{2+} _____

20. Cl^- _____

21. O_2 _____

22. P_4 _____

23. Na in Na_2S _____

24. S in H_2S _____

25. Ca^{2+} _____

26. C in CN^- _____

27. H in OH^- _____

28. Mn in KMnO_4 _____

29. I in $\text{Mg}(\text{IO}_3)_2$ _____

30. C in $\text{C}_2\text{O}_4^{2-}$ _____

Assigning Oxidation Numbers Answers

Tobias Bleicker

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Assigning Oxidation Numbers Answers

1. Understanding the eBook Assigning Oxidation Numbers Answers
 - The Rise of Digital Reading Assigning Oxidation Numbers Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Assigning Oxidation Numbers Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Assigning Oxidation Numbers Answers
 - User-Friendly Interface

4. Exploring eBook Recommendations from Assigning Oxidation Numbers Answers
 - Personalized Recommendations
 - Assigning Oxidation Numbers Answers User Reviews and Ratings
 - Assigning Oxidation Numbers Answers and Bestseller Lists
5. Accessing Assigning Oxidation Numbers Answers Free and Paid eBooks
 - Assigning Oxidation Numbers Answers Public Domain eBooks
 - Assigning Oxidation Numbers Answers eBook Subscription Services
 - Assigning Oxidation Numbers Answers Budget-Friendly Options
6. Navigating Assigning Oxidation Numbers Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Assigning Oxidation Numbers Answers Compatibility with Devices
 - Assigning Oxidation Numbers Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Assigning Oxidation Numbers Answers
 - Highlighting and Note-Taking Assigning Oxidation Numbers Answers
 - Interactive Elements Assigning Oxidation Numbers Answers
8. Staying Engaged with Assigning Oxidation Numbers Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Assigning Oxidation Numbers Answers
9. Balancing eBooks and Physical Books Assigning Oxidation Numbers Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Assigning Oxidation Numbers Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Assigning Oxidation Numbers Answers
 - Setting Reading Goals Assigning Oxidation Numbers Answers
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Assigning Oxidation Numbers Answers
 - Fact-Checking eBook Content of Assigning Oxidation Numbers Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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