

Atomic Structure Worksheet

Name _____

Fill in the blanks for the elements in this chart. For the purposes of this chart, round all atomic masses to the nearest whole number.

$$N = \text{mass\#} - P$$

= #P (in a neutral atom)

Element	Atomic Symbol	Atomic Number	Number of Protons	Mass Number	Number of Neutrons	Number of Electrons
Lithium-9	Li	3	3	9	6	3
Carbon-12	C	6	6	12	6	6
Chlorine-35	Cl	17	17	35	18	17
Silver-108	Ag	47	47	108	61	47
Lead-207	Pb	82	82	207	125	82
Calcium-40	Ca	20	20	40	20	20
Potassium-39	K	19	19	39	20	19
Radium-226	Ra	88	88	226	138	88
Aluminum-27	Al	13	13	27	14	13
Krypton-84	Kr	36	36	84	48	36
Iron-56	Fe	26	26	56	30	26
Bromine-80	Br	35	35	80	45	35
Element	Atomic Symbol	Atomic Number	Number of Protons	Atomic Mass	Number of Neutrons	Number of Electrons

- ① Sodium - 23
- ② Na
- ③ P + N

same

Atomic Structure Unit Regents Review Answers

R Bogdan



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