

# Atomic Structure

Name : \_\_\_\_\_ Date : \_\_\_\_\_

**Part A:** Use your knowledge of atomic structure to fill in the blanks.

1. The atomic number tells the number of positively charged \_\_\_\_\_ in the nucleus of an atom. The atom is \_\_\_\_\_ because the number of protons is also the number of \_\_\_\_\_ charged \_\_\_\_\_ in the atom.
2. The mass number tells the total number of \_\_\_\_\_ and \_\_\_\_\_ in the nucleus of an atom. These particles are collectively called \_\_\_\_\_ since both are located in the nucleus.
3. Isotopes are atoms of the same elements with different numbers of \_\_\_\_\_ , which results in different \_\_\_\_\_ numbers.

**Part A:** Fill in the blanks for elements in this chart. For the purpose of this chart, round the atomic masses to the nearest whole number.

Element	Number of Protons	Number of Neutrons	Number of Electrons	Atomic Mass	Atomic Number
Beryllium					
Nitrogen					
Silicon					
Argon					
Titanium					
Copper					
Bromine					
Zirconium					
Indium					
Barium					

# Atomic Structure Fill In The Blanks Answers

**Jessica J Manson**



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