

Use your periodic table to fill in the missing spaces below. Consider all atoms to be stable.

| Element | Symbol | Atomic Number | Mass Number | Protons | Neutrons | Electrons |
|-----------|--------|---------------|-------------|---------|----------|-----------|
| Carbon | | 6 | 12 | | | |
| | O | 8 | | | 8 | |
| Potassium | | | | 19 | 20 | |
| | Au | | | | | |
| Tin | Sn | 50 | | | 69 | |
| Zinc | | | 64 | 30 | | |
| Cobalt | Co | 27 | | | 32 | |
| Boron | | | | 5 | 6 | |
| | Fe | | | | | |

Use your periodic table to complete the table below. The first one has been done for you.

| Element | Atomic Number | Mass Number | Protons | Neutrons | Electrons | Symbol |
|----------|---------------|-------------|---------|----------|-----------|-----------|
| Iodine | 53 | 127 | 53 | 74 | 54 | I^{-1} |
| | | 35 | 17 | | | Cl^{-1} |
| Hydrogen | | 1 | 1 | | 0 | |
| | 35 | | | 45 | 36 | |
| | 5 | 11 | | | | B^{+3} |
| Oxygen | | 16 | 8 | | 10 | |
| Nitrogen | | | 7 | 7 | 10 | |
| Aluminum | | 27 | | | 10 | |
| | | 56 | 26 | | 24 | |
| | 29 | | | 34 | | Cu^{+1} |
| | | 63 | 29 | | 27 | |
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Atoms And Ions Worksheet Answer Key

H Kauffman



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