

Name _____

Date _____

Unit 2: Atomic Structure – Intensive Review

Atomic Models

1. What was concluded about the structure of the atom as the result of the gold foil experiment?

(1) A positively charged nucleus is surrounded by positively charged particles.

(2) A positively charged nucleus is surrounded by mostly empty space.

(3) A negatively charged nucleus is surrounded by positively charged particles.

(4) A negatively charged nucleus is surrounded by mostly empty space.

7. According to the modern model of the atom, the nucleus of an atom is surrounded by one or more

(1) electrons

(2) neutrons

(3) positrons

(4) protons

8. Which subatomic particles are located in the nucleus of a carbon atom?

(1) protons

(2) neutrons

(3) protons and neutrons

(4) protons and electrons

9. Which part of a helium atom is positively charged?

(1) electron

(2) nucleus

(3) neutron

(4) orbital

10. Answer the questions, based on your knowledge of atomic structure.

(a) Which particles have approximately the same mass?

Atomic Theory Unit Regents Review Answer Key

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