

Atomic Radiation and Half-Life

2000 pts

2000-1-1-10

Read the entire laboratory investigation and the relevant pages of your textbook, discussing it with your lab group.

1. Explain half-life, activity, radiations, isotopes, and radioactive decay.
2. The first part of the report investigates energy levels and radioactive decay.
3. Complete the following table:

TABLE 1. RADIOISOTOPES					
Isotope	Atomic Number	Atomic Mass	Number of Protons	Number of Neutrons	Number of Electrons
$^{131}_{53}\text{I}$					
$^{137}_{55}\text{Cs}$					
$^{132}_{54}\text{Xe}$					
$^{131}_{52}\text{Te}$					
$^{137}_{56}\text{Ba}$					

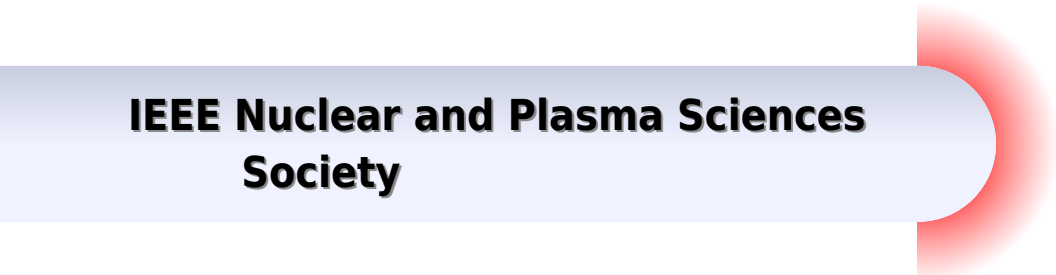
4. What is the most accurate the modeling of the isotope?
5. What is the difference between gamma radiation and alpha radiation?

APPENDIX B: RADIATION AND HALF-LIFE LAB

Radioisotopes undergo the spontaneous change in their nuclei and become other elements as some of nucleons from the nucleus of the isotope move from nucleus to nucleus without changing number and their nature, with half-life of 1 year as The Radioactive Decay, the Alpha, and other kinds of radiation. The purpose of this laboratory activity is to explore the different types of radiation, half-life, and the associated nature of radioactivity.

Atomic Radiation And Half Life Lab Answer Key

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Government Reports Announcements ,1974-07 **22nd Nuclear Science Symposium** IEEE Nuclear and Plasma Sciences Society,1976 *Resources in Education* ,1997 Radioecology U.S. Atomic Energy Commission. Division of Biology and Medicine,1963 **Freshwater and Terrestrial Radioecology** Alfred W. Klement,Vincent Schultz,1980
Annual Report Argonne National Laboratory. Chemistry Division,1970-07

Unveiling the Power of Verbal Beauty: An Psychological Sojourn through **Atomic Radiation And Half Life Lab Answer Key**

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