

Use the periodic table and your knowledge of periodic trends to answer the following questions.

Atomic Radius

- Which atom in each pair has the larger atomic radius? Circle your answer.
 a) Li or K b) Ca or Ni c) Ga or B d) O or C e) Cl or Br
 f) Be or Ba g) Si or S h) Fe or Au
- Why do atoms get smaller as you move left to right in a period? Larger as you move down a group?
- Arrange the following in order of increasing atomic radius.
 Mg, Na, P, Si and Ar _____

Ionic Radius

- Explain the relationship between the relative size of an ion to its neutral atom and the charge on the ions.
- Which particle has the larger radius in each atom/ion pair?
 a) Na, Na⁺ b) S, S²⁻ c) I, I⁻ d) Al, Al³⁺ e) Cl, Cl⁻

Ionization Energy

- Which element in each pair has the larger ionization energy?
 a) Na or O b) Be or Ba c) Ar or F d) Cu or Ra e) I or Ne
 f) K or V g) Ca or Fr h) W or Se
- Arrange the following groups of elements in order of increasing ionization energy.
 a) Be, Mg, Sr _____ b) Bi, Cs, Ba _____ c) Na, Al, S _____

Electronegativity

- Which element in each pair has a higher electronegativity value?
 a) Cl, F b) C, N c) Mg, Ne d) As, Ca e) Fr, Cs

Metallic Character

- Arrange the elements from least to most metallic: Sodium, barium, cesium, magnesium, calcium

Electron Affinity

- Write the chemical equation for the addition of an electron to a chlorine atom.
- Arrange the following atoms in order of **increasing** electron affinity: Br, Sb, I, Te, Cl

Atomic And Ionic Radii Worksheet 20 Answers

**Robert C. Smoot, Smoot, Richard G.
Smith, Jack Price**



Atomic And Ionic Radii Worksheet 20 Answers:

Class 10th Science Worksheet , This book is as per the guidelines syllabus and marking scheme issued by CBSE for Class X The salient features of this workbook are The questions in the this book have been so designed that complete syllabus is covered This book help students to identify their weak areas and improve them Additional it will help students gain confidence The questions in the book are of varying difficulty level and will help students evaluate their reasoning analysis and understanding of the subject matter *Progressive Science Class IX* Chandan Sukumar Sengupta, This hand book is meant for students having a plan for preparing Pre Medical Board Examinations and also a plan for optng competitive examinations like NEET BDS and other such entrance examinations There will be sa series of such publications which are advanced for covering different content areas of the study These are merely a reparatory study meant primarily for equipping an individual for the forthcoming challenges Contents are designed on the basis of the recommendations made by the Curriculum Framework Proposal of NCERT for Students aspiring for National Entrance Test meant for seeking admission in Under Graduate Medical Institutions There are twn such volume for clearing the fundamental concepts of Science related doubts This book has been published with all reasonable efforts taken to make the material error free after the consent of the author No part of this book shall be used reproduced in any manner whatsoever without written permission from the author except in the case of brief quotations embodied in critical articles and reviews This workbook is meant for students having eagerness for improving in later course of study in the field of science and technology It will also expose an individual to some higher challenges of studies **Merrill Chemistry** Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998

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