

# ATOMIC STRUCTURE QUIZ



**Multiple Choice:** Read the question carefully and select the choice that best answers the question. 2 pt each

1.) The charge of a proton is...

- a.) Negative
- b.) Positive
- c.) Neutral
- d.) Unknown

2.) What does the atomic number tell you?

- a.) Number of neutrons
- b.) The size of the atom
- c.) The mass of the atom
- d.) The number of protons

3.) How can you find the mass of an isotope?

- a.) Adding protons and neutrons
- b.) Adding protons and electrons
- c.) Subtracting protons and neutrons
- d.) Subtracting electrons and neutrons

4.) What is the Atomic Mass of Lithium?

- a.) 6.94
- b.) 3
- c.) 4
- d.) 6.023

5.) The charge of an electron is...

- a.) Negative
- b.) Positive
- c.) Neutral
- d.) Unknown

6.) In a Bohr model, how many electrons can go in the first orbital?

- a.) 8
- b.) 16
- c.) 2
- d.) 3

7.) What is the element symbol for Radon?

- a.) Rb
- b.) Ra
- c.) Rn
- d.) Ru

8.) Which subatomic particle does not have a mass?

- a.) Neutr
- b.) Proton
- c.) Neutron
- d.) Electron

9.) In a neutral atom, the number electrons equals the number of...

- a.) Neutrons
- b.) Protons
- c.) Mass
- d.) Cannot Determine

10.) What is the name of the element with the symbol, Hg

- a.) Mercury
- b.) Bromine
- c.) Iodine
- d.) Helium

# Atomic Structure Test Questions

**Rob Ritchie**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## Atomic Structure Test Questions:

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