

Reaction Rates Test #10

Name: _____

____K ____T ____C ____A

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(Product)

10 10 10 10

Multiple Choice: Circle the most correct answer. (10K)

1. Which one of the following is not a valid expression for the rate of the reaction below?



- A) $-\frac{1}{7} \frac{\Delta[\text{O}_2]}{\Delta t}$ B) $\frac{1}{4} \frac{\Delta[\text{NO}_2]}{\Delta t}$ C) $\frac{1}{6} \frac{\Delta[\text{H}_2\text{O}]}{\Delta t}$ D) $-\frac{1}{4} \frac{\Delta[\text{NH}_3]}{\Delta t}$

E) All of the above are valid expressions of the reaction rate

2. The activation energies for each of two reactions were found to be (a) 24 kJ and (b) 53 kJ. If the temperatures of both reactions are identical, which has the greater rate of reaction?

- A) Reaction a
B) Reaction b
C) The two reactions have the same reaction rate

3. Which of the following is true?

- A) If we know that a reaction is an elementary reaction, then we know its rate law.
B) The rate-determining step of a reaction is the rate of the fastest elementary step of its mechanism.
C) Since intermediate compounds can be formed, the chemical equations for the elementary reactions in a multistep mechanism do not always have to add to give the chemical equation of the overall process.
D) In a reaction mechanism, an intermediate is identical to an activated complex.
E) All of the above statements are true

4. Of the following, all are valid units for a reaction rate except _____

- A) mol/L B) M/s C) mol/L·s D) 0/s E) mol/L·h

5. For an open system, which of the following reactions could the reaction rate be determined by

Change in mass of the system

Change in time

- I. $2\text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2\text{H}_2\text{O}(\text{g})$
 II. $\text{CaCO}_3(\text{s}) \rightarrow \text{CaO}(\text{s}) + \text{CO}_2(\text{g})$
 III. $\text{CaO}(\text{s}) + \text{SiO}_2(\text{g}) \rightarrow \text{CaSiO}_3(\text{s})$
 IV. $\text{AgNO}_3(\text{aq}) + \text{NaCl}(\text{aq}) \rightarrow \text{NaNO}_3(\text{aq}) + \text{AgCl}(\text{s})$

Assessment Answers Reaction Rates

Alejandro Salcido

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

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