

BrainPOP: Atomic Model

Using _____ means, scientists have been able to construct different models of the _____. The atomic model has taken years and years to build and people still tweak it now and again, hopefully making it more accurate.

1802- At the beginning of the 19th century, Englishman _____ built an atomic model called the "_____."

According to his theory, each type of _____ is made of only one type of _____. Also each type of atom could be put together into a group called an _____ - and every atom of a given element was _____. For example, gold atoms make _____ and iron atoms make _____, and so on. Dalton also believed that atoms could not be _____, divided into smaller parts, or _____.

1897- About 100 years after Dalton, British scientists _____ discovered the _____. Thompson stated that negatively charged electrons must be stuck all throughout a positively charged area. Think of it like the atom is cookie dough and the _____ are chocolate chips. The discovery of electrons was a big deal. But it left some big questions- what were the positive particles holding those electrons in place?

1909- In 1909, a New Zealand born British scientist named _____ and his team set out to answer some of these questions.

After performing in depth experiments using gold foil, Rutherford determined that a gold atom must have some very small _____ charged mass. He named this mass the _____. He called the positively charged nucleus particles _____, and said that the _____ were scattered in _____ around the nucleus.

James Chadwick, a student of Rutherford's, found new particles in the nucleus that weren't affected by an electric field at all. Chadwick called these uncharged particles _____.

Also in the early 20th century, scientists discovered that electrons are arranged in energy levels. The lower energy levels are closer to the _____ and can hold just a few _____.

The higher energy levels are farther from the nucleus and can hold more _____. Electrons are so small and _____ that their energy levels are not neat planet-like orbits around the nucleus. Instead, they exist in a region called the electron cloud. The atomic model is constantly evolving as scientists discover more and more about what makes the universe tick.

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AN Whitehead



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