

Name: _____

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Formulas and Covalent Bonds Homework

Correctly fill in the box below.

METALS _____ electrons and carry a _____ charge.
(Group ____ is the most willing to lose electrons)

NONMETALS _____ electrons and carry a _____ charge.
(Group ____ is the most willing to gain electrons)

Write the correct formula for the compounds that each of the following pairs of ions will make.

H^+ F^- _____

NH_4^+ O^{2-} _____

Mg^{+2} Cl^- _____

Cu^{+2} I^- _____

Ba^{+2} CN^- _____

Ca^{+2} NO_3^- _____

NH_4^+ Cl^- _____

Sr^{+2} F^- _____

Mg^{+2} O^{2-} _____

Fe^{+2} Br^- _____

Ca^{+2} OH^- _____

Ba^{+2} NO_3^- _____

Al^{+3} SO_4^{2-} _____

Cr^{+3} O^{2-} _____

Fe^{+3} O^{2-} _____

Al^{+3} NO_3^- _____

Ionic bonds are always formed between a _____ and a _____.

Covalent bonds are ONLY formed between _____.

Decide whether the following compounds contain ionic (I) or covalent (C) bonds:

HgO _____

CCl_4 _____

NaCl _____

AgI _____

SiO_2 _____

CO _____

HgO _____

CCl_4 _____

NaCl _____

Bonding Compound Crosswords Answers

S Ben Porath



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