

Molecular Structure

Instructions: Fill in each section of this worksheet and submit it to your instructor for grading.

Review Questions

1. Write a statement explaining which characteristics of an atom determine the VSEPR shape of a molecule (hint: look at the VSEPR geometry chart).
VSEPR shows the shape of a molecule and the arrangement of its electron pairs and the atom, it predicts the lone pairs and it predicts shape.
2. How does the force between the atoms impact the shape of a molecule?
The force between the atom determines the shape.

Activity: First row is done for you as a model.

Compound	Draw Lewis Structure	Use VSEPR Chart to determine:	Determine the structures of the molecule	Draw the VSEPR model
H ₂ O		Total Domains: 4 Bonded Atoms: 2 Lone Pairs: 2	Molecular Shape: Bent Electron Geometry: Tetrahedral	
SiH ₄		Total Domains: 4 Bonded Atoms: 4 Lone Pairs: 0	Molecular Shape: tetrahedral Electron Geometry: tetrahedral	
Br ₂		Total Domains: 1 Bonded Atoms: 1 Lone Pairs: 0	Molecular Shape: linear Electron Geometry: linear	

Bonding And Molecular Structure Worksheet Answers Year 9

Lingsheng Yao



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Exploration of the Structure of Atom Chandan Sengupta, First Publication April 2019 Revised Publication October 2022 Second Revised Edition July 2023 Third Revised Edition December 2024 Published by Chandan Senguta Printed by IECIT Printing and ublication Services Resource Centre Arabinda Nagar Bankura 722101 W B India Topics Covered Atoms and Molecules Structure of Atom Bonding Mechanism and Chemical Reactions Mechanism of Bonding This book is suitable for students of Class 9 to Class 11 Students aspiring for Pre Medical Entrance Examination can also get adequate support Additional Hard Copies can also be obtained from Chandan Sukumar Sengupta Arabinda Nagar Bankura 722101 WB Write to Us for more materials

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