

Changes in Bonding

Why haven't we progressed on regarding all of the elements we've met? Because most substances are compounds. They have elements and stable enough to a net gain. Instead they must make other substances to become compounds. Why?

Shared Bonds—electrons shared to achieve a full set of valence electrons. This is usually 8 (what are the exceptions). Atoms can make stable molecules that meet this.

To fulfill the octet rule atoms will lose, gain, or share electrons. Atoms will gain electrons to achieve the full octet; usually will lose electrons; some atoms will gain electrons.

Why? Atoms have outer electrons that don't shield well. They have low energy electrons that have low energy electrons for the nucleus. (Electrons) have outer electrons that are not well shielded from the nucleus.

Ionization Energy

Ionization energy is the energy needed to remove an electron from an atom. It is the energy needed to remove an electron from an atom. It is the energy needed to remove an electron from an atom.



Charge—positive ions (cations).
 Loss of electrons.
 Atoms—negative ions (anions).
 Gain of electrons.

How the ionization energy changes with increasing atomic number.

How the ionization energy changes with increasing atomic number.

Why does a metal like iron atoms lose electrons?

Ionization Energy—Energy of electrons

Ionization energy is the energy needed to remove an electron from an atom.



How does the ionization energy change with increasing atomic number?

It is low for metals.

How does the ionization energy change with increasing atomic number?

How does the ionization energy change with increasing atomic number?

It is high for non-metals.

How does the ionization energy change with increasing atomic number?



It is low for metals.



Sodium Chloride—Tablet salt.

Why does a metal like iron atoms lose electrons?

Why does a non-metal like chlorine atoms gain electrons?

Na^{+} and Cl^{-} _____ Fe^{2+} and O^{2-} _____

Fe^{3+} and O^{2-} _____ Al^{3+} and S^{2-} _____

Fe and Cl^{-} _____ Al and O^{2-} _____

Fe and Cl^{-} _____ Al and O^{2-} _____

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Jicheng Xie



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Unveiling the Energy of Verbal Artistry: An Emotional Sojourn through **Bonding Answer Key For Stephen Murray**

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