Name:	Date:	Class period

Unit Test: Chemical Bonding (A) ANSWER KEY

100

Part I: Matching (2 points each) Please match the term with the correct definition. There will be extra choices and some choices may be used more than once.

A	Electronegativity	F. Molecular Formula	K. Crystal
	Covalent bonding	G. Linear	L. Pyramid
	Ductile	H. Non-polar	M. Polar
D.	Agueous	I. Solid	N. VSEPR
E.	Tetrahedral	 J. Trigonal bipyramidal 	O. Malleable
	Octet Rule	Q. Ionic bonding	R. Empirical Formula

- 1. _K_ A 3-D, repeating pattern of + and ions, formed by ionic compound
- Zendency for an atom to attract electrons when chemically bonded to another atom
- B A sharing of electrons
- _P__ Atoms will gain or lose enough electrons in order to become isoelectronic with a noble gas
- Q A transfer of electrons
- 6. __R__ A chemical formula that is arranged in the smallest whole number ratio
- 7. __D_ The term that means dissolved in water
- 8. __F__ A chemical formula that describes the makeup of a single molecule
- The shape (geometry) that is always an exception to the octet rule
- 10. __M_ A bond where electrons are shared unequally between atoms
- 11. L One of the shapes (geometries) that will always be polar
- 12. __H__ A bond where electrons are shared equally between atoms

Part II: Multiple Choice (2 points each) Please circle the <u>most correct</u> answer. Write your answer on the given lines.

- 1. E A triple covalent bond shares electrons between atoms.
 a. two
 b. three
 c. four
 d. five
 e. six
 f. Unable to tell

Bonding Test Answer Key

Ray Holder

Bonding Test Answer Key:

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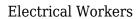
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