

CHAPTER 6 REVIEW*Chemical Bonding***MIXED REVIEW**

SHORT ANSWER Answer the following questions in the space provided.

1. Name the type of energy that is a measure of strength for each of the following types of bonds:

_____ lattice energy _____	a. ionic bond
_____ bond energy _____	b. covalent bond
_____ enthalpy of vaporization _____	c. metallic bond

2. Use the electronegativity values shown in **Figure 20**, on page 161 of the text, to determine whether each of the following bonds is nonpolar covalent, polar covalent, or ionic.

_____ ionic _____	a. H—F	_____ nonpolar covalent _____	d. H—H
_____ ionic _____	b. Na—Cl	_____ polar covalent _____	e. H—C
_____ polar covalent _____	c. H—O	_____ polar covalent _____	f. H—N

3. How is a hydrogen bond different from an ionic or covalent bond?

A hydrogen bond is a dipole-dipole attraction between a partially positive hydrogen
atom and the unshared electron pair of a strongly electronegative atom such as O,
N, or F. Unlike ionic or covalent bonds, in which electrons are given up or shared,
the hydrogen bond is a weaker attraction. Hydrogen bonds are generally
intermolecular, while ionic and covalent bonds occur between ions or atoms
respectively.

4. H₂S and H₂O have similar structures and their central atoms belong to the same group. Yet H₂S is a gas at room temperature and H₂O is a liquid. Use bonding principles to explain why this is.

Oxygen has higher electronegativity than sulfur, which creates a highly polar bond.
Increased polarity in H₂O bonds means a stronger intermolecular attraction, making
water a liquid at room temperature. Hydrogen bonding exists between water
molecules, but not between hydrogen sulfide molecules.

Covalent Bonding Chapter 8 Worksheet Answers

Sabine Zange



Covalent Bonding Chapter 8 Worksheet Answers:

Unveiling the Energy of Verbal Art: An Emotional Sojourn through **Covalent Bonding Chapter 8 Worksheet Answers**

In a global inundated with displays and the cacophony of quick transmission, the profound power and emotional resonance of verbal artistry frequently disappear in to obscurity, eclipsed by the continuous assault of noise and distractions. However, nestled within the lyrical pages of **Covalent Bonding Chapter 8 Worksheet Answers**, a interesting perform of literary brilliance that pulses with fresh emotions, lies an wonderful trip waiting to be embarked upon. Penned by a virtuoso wordsmith, that enchanting opus manuals viewers on an emotional odyssey, delicately revealing the latent potential and profound influence stuck within the complicated web of language. Within the heart-wrenching expanse of this evocative analysis, we can embark upon an introspective exploration of the book is central styles, dissect their fascinating publishing type, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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