

\_\_\_\_K \_\_\_\_T \_\_\_\_C \_\_\_\_A  
20 5 8 8

# Assessment of Learning (Product)

Multiple Choice: Circle the most correct answer. (20K)

1. The element X has three naturally occurring isotopes. The isotopic masses (amu) and % abundances of the isotopes are given in the table below. The average atomic mass of the element is \_\_\_\_\_ amu.

| Isotope         | Abundance | Mass  |
|-----------------|-----------|-------|
| $^{55}\text{X}$ | 19.61     | 52.62 |
| $^{56}\text{X}$ | 53.91     | 56.29 |
| $^{58}\text{X}$ | 26.48     | 58.21 |

- A) 33.33      B) 55.74      C) 56.11      D) 57.23      E) 56.29

2. The seven elements that occur as diatomic elements are:

- A)  $\text{H}_2$ ,  $\text{N}_2$ ,  $\text{O}_2$ ,  $\text{He}_2$ ,  $\text{Ne}_2$ ,  $\text{Cl}_2$ ,  $\text{Br}_2$       B)  $\text{H}_2$ ,  $\text{N}_2$ ,  $\text{O}_2$ ,  $\text{He}_2$ ,  $\text{Ne}_2$ ,  $\text{C}_2$ ,  $\text{Na}_2$   
C)  $\text{H}_2$ ,  $\text{N}_2$ ,  $\text{O}_2$ ,  $\text{F}_2$ ,  $\text{Cl}_2$ ,  $\text{Br}_2$ ,  $\text{I}_2$       D)  $\text{Fe}_2$ ,  $\text{Ru}_2$ ,  $\text{O}_2$ ,  $\text{He}_2$ ,  $\text{Ne}_2$ ,  $\text{C}_2$ ,  $\text{Br}_2$

3. What is the total number of electrons in a  $\text{Mg}^{2+}$  ion?

- A) 10      B) 14      C) 12      D) 24

4. Atoms of which element have the greatest tendency to gain electrons?

- A) bromine      B) fluorine      C) chlorine      D) iodine

5. Use the periodic table to predict which element has the largest ionization energy.

- A) H      B) Kr      C) Fr      D) Rn      E) Rh (# 45)

6. As you move across a period, from left to right, the size of the atoms of each element:

- A) increases      B) decreases      C) stays the same      D) none of the above

7. The least electronegative elements are the:

- A) Alkali metals      B) Metalloids      C) Halogens      D) Noble gases

8. In which one of these parts of the periodic table are transition metals located?

- A) in groups 1 & 2 at the left side of the table  
B) in group 18 at the right side of the table  
C) in groups 3-12 across the middle of the table  
D) in the lanthanoid and actinoid series

9. An atom has an atomic number of 16 and a mass number of 32. The number of electrons in the third orbit of the atom is:

- A) 2      B) 3      C) 5      D) 6      E) 8

# Core Teaching Resources Chemistry The Periodic Table Test

**New York (State). Army National  
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## **Core Teaching Resources Chemistry The Periodic Table Test:**

*Resources in Education*, 1998      **Teaching and Learning Online** Franklin S. Allaire, Jennifer E. Killham, 2023-01-01

Science is unique among the disciplines since it is inherently hands on. However, the hands on nature of science instruction also makes it uniquely challenging when teaching in virtual environments. How do we as science teachers deliver high quality experiences to secondary students in an online environment that leads to age grade level appropriate science content knowledge and literacy but also collaborative experiences in the inquiry process and the nature of science? The expansion of online environments for education poses logistical and pedagogical challenges for early childhood and elementary science teachers and early learners. Despite digital media becoming more available and ubiquitous and increases in online spaces for teaching and learning (Killham et al 2014, Wong et al 2018), PreK-12 teachers consistently report feeling underprepared or overwhelmed by online learning environments (Molnar et al 2021, Seaman et al 2018). This is coupled with persistent challenges related to elementary teachers' lack of confidence and low science teaching self-efficacy (Brigido, Borrachero, Bermejo, Gunning, Duran, Duran 2004). Ideal for undergraduate teacher candidates, graduate students, teacher educators, classroom teachers, parents, and administrators, this book addresses why and how teachers use online environments to teach science content and work with elementary students through a research-based foundation.

**Chemistry** Richard Post, Chad Snyder, Clifford C. Houk, 2020-09-16

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**Biology** Steven D. Garber, 2020-09-16

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**New York Army National Guard, Education Services Handbook** New York (State). Army National Guard,1986

**What Teachers Really Need to Know About Formative Assessment** Laura Greenstein,2010-06-28 Explains how to make formative assessments a seamless and natural part of the teaching process and provides assessment strategies that can be used before during and after instruction to learning

**The Software Encyclopedia 2000** Bowker Editorial Staff,2000-05

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