

ACIDS, BASES, & THE pH SCALE

Name: **Answer Key**

Date: _____

Pd: _____

PART 1: ACIDS AND BASES

Comparing Acids and Bases	
Acids	Bases (Alkalines)
• Sour taste, leaves burning sensation • Increase hydrogen ions (H^+) in solution • Example: HCl $H^+ + Cl^-$ • Turns litmus (pH) paper to red	• Sharp bitter taste, feels slippery to touch • Decrease hydrogen ions (H^+) in solution (Increase hydroxide ions (OH^-) in solution) • Example: NaOH $Na^+ + OH^-$ • Turns litmus (pH) paper to blue

PART 2: THE pH SCALE

- The **pH Scale** is a **measurement** of how acidic or how basic a solution is.
- The scale goes from **0** to **14**.
- **Compounds** are **acidic** if they have a pH **lower** than **7**.
- Compounds with a pH **higher** than **7** are **basic** or **alkaline**.
- Compounds with a pH **of 7**, are **neutral**.

0 1 2 3 4 5 6 7 8 9 10 11 12 13 14

Increased concentration of **H^+** ions
Decreased concentration of **OH^-** ions

Decreased concentration of **H^+** ions
Increased concentration of **OH^-** ions

pH below 7 = **acids**

pH at 7 = **neutral**
(**pure water**)

pH above 7 = **bases**

PART 3: BUFFERS

- A **buffer** is a **weak acid** or **base** that **prevents** sudden **changes** in pH.
 - Maintains **homeostasis**
- A buffer works by **binding** to H^+ ions **releasing** H^+ ions when the solution pH changes.
- Buffers **bind** to H^+ ions when the **concentration** of H^+ ions in solution **increases**.
- **NaOH** **$Na^+ + OH^-$** (The OH^- will bind to H^+ ions to form H_2O – **neutralization**)
- Buffers **release** H^+ ions when the **concentration** of H^+ ions in solution **decreases**.
HCl **$H^+ + Cl^-$**

Acids And Bases Answer Key

Xiang Xie



Acids And Bases Answer Key:

Reviewing **Acids And Bases Answer Key**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Acids And Bases Answer Key**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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