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FOCUS 2

MOLECULES

2A.2 (a) 5; (b) 8; (c) 12; (d) 7

2A.4 (a) $[\text{Ar}] 3d^1$; (b) $[\text{Ar}] 3d^5$; (c) $[\text{Kr}] 4d^{10}$; (d) $[\text{Xe}]$

2A.6 (a) $[\text{Kr}]$; (b) $[\text{Xe}] 4f^{14}5d^5$; (c) $[\text{Xe}]$; (d) $[\text{Ar}]$

2A.8 (a) Ca: $[\text{Ar}] 4s^2$, Ti^{2+} : $[\text{Ar}] 3d^2$; V^{3+} : $[\text{Ar}] 3d^2$ In the d block, the energies of the $n-1$ d-orbitals lie below those of the ns-orbitals. Therefore, when V and Ti form ions they lose their 3s electrons before losing their 4d electrons. (b) Ca: no unpaired electrons; Ti^{2+} : two unpaired electrons; V^{3+} : two unpaired electrons. (c) Ti^{2+} : $[\text{Ar}] 3d^1$ no neutral atom has this electron configuration.

2A.10 (a) Au^{3+} ; (b) Os^{3+} ; (c) I^{3+} ; (d) As^{3+}

2A.12 (a) Cr^{2+} ; (b) Ag^{2+} ; (c) Zn^{2+} ; (d) Pb^{2+}

2A.14 (a) 4d; (b) 3p; (c) 5p; (d) 4s

2A.16 (a) +1; (b) -2; (c) +2; (d) -3; (e) -1

Atkins 7th Edition Solutions Manual

Peter William Atkins



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Student Solutions Manual to Accompany Atkins' Physical Chemistry 11th Edition Peter Bolgar, Vladimiras Oleinikovas, Aimee North, Haydn Lloyd, James Keeler, Stephanie Smith, 2018 The Student Solutions Manual to accompany Atkins Physical Chemistry 11th Edition provides full worked solutions to the a exercises and the odd numbered discussion questions and problems presented in the parent book The manual is intended for students and provides helpful comments and friendly advice to aid understanding

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Table of Contents Atkins 7th Edition Solutions Manual

1. Understanding the eBook Atkins 7th Edition Solutions Manual
 - The Rise of Digital Reading Atkins 7th Edition Solutions Manual
 - Advantages of eBooks Over Traditional Books
2. Identifying Atkins 7th Edition Solutions Manual
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atkins 7th Edition Solutions Manual
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atkins 7th Edition Solutions Manual
 - Personalized Recommendations
 - Atkins 7th Edition Solutions Manual User Reviews and Ratings
 - Atkins 7th Edition Solutions Manual and Bestseller Lists
5. Accessing Atkins 7th Edition Solutions Manual Free and Paid eBooks
 - Atkins 7th Edition Solutions Manual Public Domain eBooks
 - Atkins 7th Edition Solutions Manual eBook Subscription Services
 - Atkins 7th Edition Solutions Manual Budget-Friendly Options
6. Navigating Atkins 7th Edition Solutions Manual eBook Formats
 - ePub, PDF, MOBI, and More
 - Atkins 7th Edition Solutions Manual Compatibility with Devices
 - Atkins 7th Edition Solutions Manual Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Atkins 7th Edition Solutions Manual
 - Highlighting and Note-Taking Atkins 7th Edition Solutions Manual
 - Interactive Elements Atkins 7th Edition Solutions Manual

8. Staying Engaged with Atkins 7th Edition Solutions Manual
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atkins 7th Edition Solutions Manual
9. Balancing eBooks and Physical Books Atkins 7th Edition Solutions Manual
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atkins 7th Edition Solutions Manual
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Atkins 7th Edition Solutions Manual
 - Setting Reading Goals Atkins 7th Edition Solutions Manual
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Atkins 7th Edition Solutions Manual
 - Fact-Checking eBook Content of Atkins 7th Edition Solutions Manual
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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