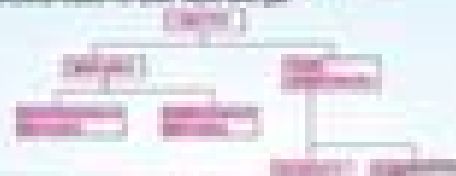


1. CLASSIFICATION OF MATTER

- Matter can exist in three physical states: solid, liquid and gas.



- Various properties of matter are used to differentiate between the two.
- Matter is classified into pure and impure.
- Homogeneous mixture: composition is uniform throughout.
- Heterogeneous mixture: composition is not uniform.
- Colloid: particles of one substance are dispersed in another.
- Suspension: particles of one substance are dispersed in another.

2. PROPERTIES OF MATTER AND THEIR MEASUREMENT

- Properties of matter are physical and chemical properties.
 - Physical properties are those which can be measured without changing the identity of the substance.
 - Chemical properties are those which can be measured by changing the identity of the substance.
 - Colour, odour, melting point, density, etc. are some physical properties.
 - Combustion, corrosion, etc. are some chemical properties.
 - Quantitative measurement of physical properties is necessary for scientific study.
 - Properties of matter are measured in SI units.
 - Some important quantities and their units are:
- | Quantity | SI Unit |
|------------------|------------------------------------|
| Mass (g) | Kilogram (kg) |
| Volume (mL) | Cubic decimetre (dm ³) |
| Temperature (°C) | Kelvin (K) |

Some important units and their conversion

1 kg = 1000 g	1 g = 1000 mg	1 kg = 1000 g	1 kg = 1000 g	1 kg = 1000 g
1 kg = 1000 g	1 g = 1000 mg	1 kg = 1000 g	1 kg = 1000 g	1 kg = 1000 g

Some important units

- 1 kg = 1000 g, 1 g = 1000 mg, 1 kg = 1000 g, 1 kg = 1000 g, 1 kg = 1000 g

3. UNCERTAINTY IN MEASUREMENT

- Measurement is a comparison of a quantity with a standard unit.
- Measurement is always uncertain.
- Measurement is always uncertain.

- Significant figures are those digits which are known with certainty plus one additional uncertain digit.

Rules for determining the number of significant figures

- Non-zero digits are significant.
- Zeros between two non-zero digits are significant.
- Zeros at the end or right of a number are significant if they are on the right side of a decimal point.
- Leading zeros are not significant.
- In a number written in scientific notation, all digits are significant.
- Exact values are the values of numbers which are not measured.
- Round-off is the process of converting a number to the required number of significant figures.
- In addition and subtraction of significant figures, the result should have the same number of significant figures as the least number of significant figures in the original numbers.
- In multiplication and division of significant figures, the result should have the same number of significant figures as the least number of significant figures in the original numbers.

Aakash Chemistry Lab Manual

S. K. Sinha



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