# Chapter 2: Elements are the Building Blocks of Matter Study Guide "Without pressure, the work doesn't get done at all"

William Sarovan (1902-1931)

#### Steps for studying for the Ch 2 test

- Re-sead your notes and use your Cornell-questions to quiz yourself.
- Re-read pages 40 73 of your textbook.
- Be able to define the key vocabulary from the chapter.
- Review your Reading Checks and Work Book, making sure all is complete.
- Be able to answer the questions below. They are a guide for your studying.
- Complete the online quizzes at http://www.bcscience.com/bc9/

metalloids

#### Chapter 2 VOCABULARY (pages 40 -73)\*\*Make vocab cards and quiz yourself if you need to!

elements	<ul> <li>atomic</li> </ul>	<ul> <li>period</li> </ul>	<ul> <li>electron</li> </ul>
compounds	namber	<ul> <li>chemical</li> </ul>	shells
chemical	<ul> <li>atomic mass</li> </ul>	family	<ul> <li>valence</li> </ul>
symbol	<ul> <li>actornic mass</li> </ul>	alkali metals	electrons
chemical	vanit	<ul> <li>allcaline</li> </ul>	<ul> <li>valence shell</li> </ul>
properties	<ul> <li>ion charge</li> </ul>	earth metals	<ul> <li>noble gas</li> </ul>
metals	• ion	<ul> <li>halogens</li> </ul>	stability
non-metals	<ul> <li>multiple ion</li> </ul>	<ul> <li>noble gases</li> </ul>	
periodic	charge	<ul> <li>Bohr model</li> </ul>	

of the atom.

#### Section 2.1: Elements

codo be-

- How many elements are there?
- Give 2 examples of elements at each of the following states:

a) solid b) liquid c) gas

- 3 Most of the elements names are derived from what languages?
- Be familiar with the physical and chemical properties, some uses, and interesting information of the following elements.

a) Hydrogen e) Chlorine
b) Iron f) Mercury
c) Oxygen g) Silver
d) Sodium h) Silicon

5. Complete the Section 2.1 Quiz online at http://www.bcscience.com/bc9/

#### Section 2.2: The Periodic Table and Chemical Properties

- a) Who is Dmitn Mendeleev?
  - b) Describe his early work on organizing the elements.
  - c) Describe Mendeleev's special insight in the organization of elements.
- What does the Periodic Table show for each element?
- 3. a). Differentiate atomic number from atomic mass and choose a specific element for example.
  - b). Describe the pattern of each on the Penodic Table.
- Differentiate atoms from ions.
- a) What are the 2 types of ions?

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William Edmund W. Collins

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answers We ll tell you everything you need to know about 145 great careers from accountant to zoologist including Day to day activities and responsibilities of each career How to plan ahead while you re still in high school Majors to consider when you get to college Whether you ll strike it rich in more ways than one Words of wisdom from professionals in the field This book will open your eyes to a world of career possibilities and give you an idea of what to expect down each career path Even if you have no clue about what you want to be this is the book for you We ll help you identify the kinds of jobs that will help you be yourself and incorporate your interests values and skills There s no pressure for you to choose a career just yet so just relax and have fun exploring

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