

Name: \_\_\_\_\_

Date: \_\_\_\_\_

Period: \_\_\_\_\_

## Formulas and Covalent Bonds Homework

Correctly fill in the box below.

METALS \_\_\_\_\_ electrons and carry a \_\_\_\_\_ charge.  
(Group \_\_\_\_ is the most willing to lose electrons)

NONMETALS \_\_\_\_\_ electrons and carry a \_\_\_\_\_ charge.  
(Group \_\_\_\_ is the most willing to gain electrons)

Write the correct formula for the compounds that each of the following pairs of ions will make.

$\text{H}^+$   $\text{F}^-$  \_\_\_\_\_

$\text{NH}_4^+$   $\text{O}^{2-}$  \_\_\_\_\_

$\text{Mg}^{+2}$   $\text{Cl}^-$  \_\_\_\_\_

$\text{Cu}^{+2}$   $\text{I}^-$  \_\_\_\_\_

$\text{Ba}^{+2}$   $\text{CN}^-$  \_\_\_\_\_

$\text{Ca}^{+2}$   $\text{NO}_3^-$  \_\_\_\_\_

$\text{NH}_4^+$   $\text{Cl}^-$  \_\_\_\_\_

$\text{Sr}^{+2}$   $\text{F}^-$  \_\_\_\_\_

$\text{Mg}^{+2}$   $\text{O}^{2-}$  \_\_\_\_\_

$\text{Fe}^{+2}$   $\text{Br}^-$  \_\_\_\_\_

$\text{Ca}^{+2}$   $\text{OH}^-$  \_\_\_\_\_

$\text{Ba}^{+2}$   $\text{NO}_3^-$  \_\_\_\_\_

$\text{Al}^{+3}$   $\text{SO}_4^{2-}$  \_\_\_\_\_

$\text{Cr}^{+3}$   $\text{O}^{2-}$  \_\_\_\_\_

$\text{Fe}^{+3}$   $\text{O}^{2-}$  \_\_\_\_\_

$\text{Al}^{+3}$   $\text{NO}_3^-$  \_\_\_\_\_

Ionic bonds are always formed between a \_\_\_\_\_ and a \_\_\_\_\_.

Covalent bonds are ONLY formed between \_\_\_\_\_.

Decide whether the following compounds contain ionic (I) or covalent (C) bonds:

$\text{HgO}$  \_\_\_\_\_

$\text{CCl}_4$  \_\_\_\_\_

$\text{NaCl}$  \_\_\_\_\_

$\text{AgI}$  \_\_\_\_\_

$\text{SiO}_2$  \_\_\_\_\_

$\text{CO}$  \_\_\_\_\_

$\text{HgO}$  \_\_\_\_\_

$\text{CCl}_4$  \_\_\_\_\_

$\text{NaCl}$  \_\_\_\_\_

# Bonding Compound Crosswords Answers

**Rachel Sandford**



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