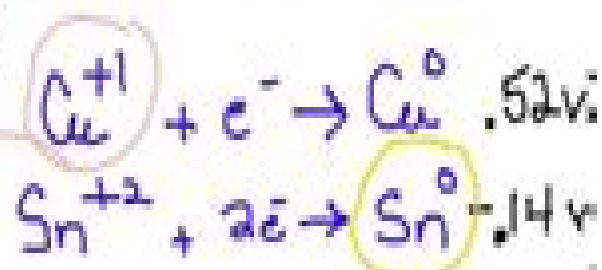
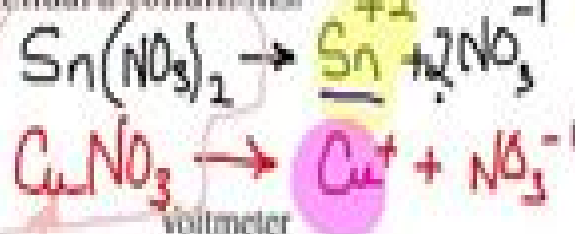


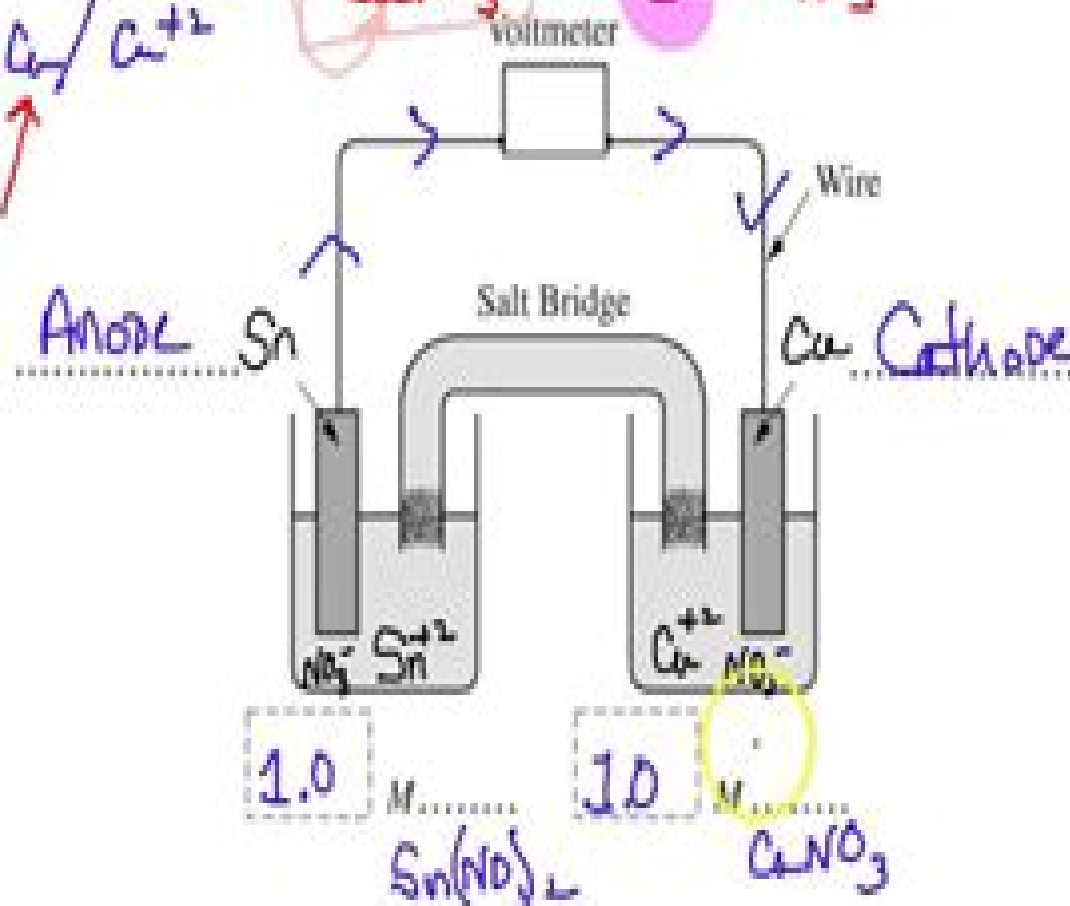
- c) Label each electrode with the appropriate metal (next to each arrow)
 d) Identify the Cathode and Anode on the dotted lines
 e) Write the correct salt in the dotted boxes underneath each half cell.
 f) Write the correct molarity if at Standard conditions.



Materials: Sn, Cu, $\text{Sn}(\text{NO}_3)_2$, CuNO_3
 Salt Bridge compound: KCl



Sn/Sn^{+2}
 Cu/Cu^{+2}



Anode
An ox =

g) oxidation half reaction: _____ $\rightarrow$ _____ v

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Jianjun Gao



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