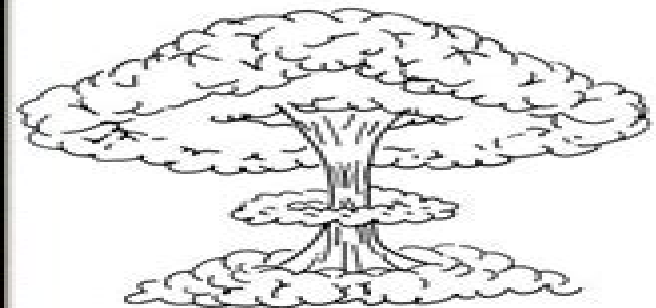


# NUCLEAR REACTIONS

## FUSION & FISSION



## Cornell Doodle Notes



**NUCLEAR REACTIONS (FUSION & FISSION)**

**DEFINITIONS:** How is energy generated from atoms?

**NOTE: SUBATOMICS:** The nucleus of an atom is composed of two subatomic particles.

**1** What particles make up the nucleus of the atom?

**2** What is an electron in an atom?

**3** What is fusion? How does it happen?

**4** What is fission? How does it happen?

**FUSION:** The process of two light nuclei combining to form a heavier nucleus, releasing energy. Example:  $2H + 3H \rightarrow 4He + 2n$ . Energy: The energy of an atom is released when the nucleus is broken apart.

**FISSION:** The process of a heavy nucleus splitting into two or more lighter nuclei, releasing energy. Example:  $^{235}U + n \rightarrow ^{141}Ba + ^{92}Kr + 3n$ . Energy: The energy of an atom is released when the nucleus is broken apart.

**NUCLEAR REACTIONS:** Chain Reaction: A series of nuclear reactions where the products of one reaction become the reactants of the next. Control: The process of controlling the rate of a nuclear reaction.



**FUSION**

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**NUCLEAR REACTIONS:** Chain Reaction: A series of nuclear reactions where the products of one reaction become the reactants of the next. Control: The process of controlling the rate of a nuclear reaction.

**E=mc<sup>2</sup>**

Energy = mass x speed of light squared

Example: The energy released in the fission of  $^{235}U$  is greater than the energy released in the fusion of  $2H$  and  $3H$ .

Energy: The energy of an atom is released when the nucleus is broken apart.

# Fission Vs Fusion And Chain Reactions Worksheet Answer Key

**F. T. Adler, H. von Halban**



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## **Table of Contents Fission Vs Fusion And Chain Reactions Worksheet Answer Key**

1. Understanding the eBook Fission Vs Fusion And Chain Reactions Worksheet Answer Key
  - The Rise of Digital Reading Fission Vs Fusion And Chain Reactions Worksheet Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Fission Vs Fusion And Chain Reactions Worksheet Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Fission Vs Fusion And Chain Reactions Worksheet Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Fission Vs Fusion And Chain Reactions Worksheet Answer Key
  - Personalized Recommendations
  - Fission Vs Fusion And Chain Reactions Worksheet Answer Key User Reviews and Ratings
  - Fission Vs Fusion And Chain Reactions Worksheet Answer Key and Bestseller Lists
5. Accessing Fission Vs Fusion And Chain Reactions Worksheet Answer Key Free and Paid eBooks
  - Fission Vs Fusion And Chain Reactions Worksheet Answer Key Public Domain eBooks
  - Fission Vs Fusion And Chain Reactions Worksheet Answer Key eBook Subscription Services
  - Fission Vs Fusion And Chain Reactions Worksheet Answer Key Budget-Friendly Options

6. Navigating Fission Vs Fusion And Chain Reactions Worksheet Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Fission Vs Fusion And Chain Reactions Worksheet Answer Key Compatibility with Devices
  - Fission Vs Fusion And Chain Reactions Worksheet Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Fission Vs Fusion And Chain Reactions Worksheet Answer Key
  - Highlighting and Note-Taking Fission Vs Fusion And Chain Reactions Worksheet Answer Key
  - Interactive Elements Fission Vs Fusion And Chain Reactions Worksheet Answer Key
8. Staying Engaged with Fission Vs Fusion And Chain Reactions Worksheet Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Fission Vs Fusion And Chain Reactions Worksheet Answer Key
9. Balancing eBooks and Physical Books Fission Vs Fusion And Chain Reactions Worksheet Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Fission Vs Fusion And Chain Reactions Worksheet Answer Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Fission Vs Fusion And Chain Reactions Worksheet Answer Key
  - Setting Reading Goals Fission Vs Fusion And Chain Reactions Worksheet Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Fission Vs Fusion And Chain Reactions Worksheet Answer Key
  - Fact-Checking eBook Content of Fission Vs Fusion And Chain Reactions Worksheet Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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