

A2 Unit F334: Chemistry of Materials

Topic 3: What's in a medicine? (date)

A study of medicines such as aspirin, their development, chemistry and synthesis, illustrating some of the features of the pharmaceutical industry. The chemical ideas in this module are:

- Molecular synthesis: acids, esters, carbonyl compounds
- Acid-base reactions
- Medicines: manufacture and testing
- IR spectroscopy and mass spectroscopy

| Topic | Alt code | Number | Assessable learning outcomes | Number of questions | OSP A2 Revision guide (Page number) | Chemical changes (Page number) | Chemical ideas (Page number) |
|-----------------------|----------|--------|--|---------------------|-------------------------------------|--------------------------------|------------------------------|
| Equilibria | | WH01 | Describe acids in terms of the Brønsted-Lowry theory as proton donors, and bases as proton acceptors | 1 | 6 | | 180 |
| | | WH02 | Identify the proton donor and proton acceptor in an acid-base reaction | 1 | 6 | | 180 |
| Bonding and structure | FL10 | WH03 | Draw and interpret simple electron-dot-and-cross diagrams to show how atoms (bond through one, covalent and three covalent bonds) | 2 | 14 | | 181-182 |
| | FL10a | WH04 | Describe a simple model of metallic bonding | | 1 | | 41 |
| | GL10 | WH05 | Use the electron-dot repulsion principle to predict and explain the shapes of simple molecules (such as CH ₄ , NH ₃ , H ₂ O and HF) and ions (such as NH ₄ ⁺) with up to six outer pairs of electrons. (any combination of bonding pairs and lone pairs (no treatment of hybridisation or molecular orbitals is expected)) | 1 | 7 | | 181-182 |

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