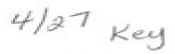
#### Colligative Properties



 As a solute is added to a solvent, what happens to the freezing point and the boiling point of the solution?

A) The freezing point decreases and the boiling point increases.

- B) The freezing point increases and the boiling point increases.
- The freezing point increases and the boiling. point decreases.
- D) The freezing point decreases and the boiling. point decreases.
- Compared to the freezing point of 1.0 M KCl(aq) at standard pressure, the freezing point of 1.0 M. CaCla(aq) at standard pressure is

B) higher

Which aqueous solution of KI freezes at the lowest реширеги писе?

A) 1 mol of KI in 500. g of water

B) 2 mol of KI in 500. g of water (%)

C) I mol of KI in 1000, g of water

D) 2 mol of KI in 1000, g of water

 A I kilogram sample of water will have the highest freezing point when it contains

A) 1 × 10<sup>21</sup> dissolved particles

(B) 1 × 1017 dissolved particles

C) 1 × 1019 dissolved particles

D) 1 × 10<sup>23</sup> dissolved particles laucs.

 A solution consists of 0.50 mole of CaCl: dissolved in 100, grams of H2O at 25°C. Compared to the boiling point and freezing point of 100. grams of H2O at standard pressure, the solution at standard pressure has

 A) a higher boiling point and a lower freezing. point

B) a lower boiling point and a lower freezing

C) a higher boiling point and a higher freezing.

 D) a lower boiling point and a higher freezing. point

6. Which solution will freeze at the lowest bisimplerature?

A) I mole of sugar in 500 g of water

B) I mole of sugar in 1,000 g of water

C) 2 moles of sugar in 1,000 g of water 2 moles of sugar in 500 g of water

7. Which solution has the highest boiling point at standard pressure?

(A))0.10 M K;PO4(30)

B) 0.10 M KCl(ac)

C) 0.10 M KNO<sub>3</sub>(a<sub>0</sub>)

D) 0.10 M K2SO4(aq)

 Which concentration of a solution of CH1OH in water has the lowest freezing point?

A) 0.001 M

C) 0.01 M

9. How do the boiling point and freezing point of a solution of water and calcium chloride at standard pressure compare to the boiling point and freezing

point of water at standard pressure?

 A) Both the freezing point and boiling point of the solution are higher.

B) The freezing point of the solution is higher and the boiling point of the solution is lower.

C) The freezing point of the solution is lower and the boiling point of the solution is higher.

 D) Both the freezing point and boiling point of the solution are lower.

10. Compared to a 2.0 M aqueous solution of NaCl at 1 atmosphere, a 3.0 M aqueous solution of NaC1 at I atmosphere has a

A) lower boiling point and a higher freezing.

B) higher boiling point and a higher freezing.

C) lower boiling point and a lower freezing provinct.

higher boiling point and a lower freezing point

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**SA Adler** 

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web purposes objectives in this lab you will be able to make atom models that show stable atoms or ions use given information about subatomic particles to a identify an element and its position on the periodic table b draw models of atoms using build an atom phet sim c determine if the model is for a neutral atom or an ion

## 1 1 build an atom phet simulation flashcards quizlet - Nov 29 2022

web what particles are located in the center of an atom click the card to flip protons and neutrons click the card to flip 1 22

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1 4 experiment 2 atomic theory chemistry libretexts - Feb 01 2023

web using zoom breakout groups you can work together with your group members to answer the questions in the atomic theory lab report this is an individual assignment so while you can work with your peers all work must be in your own words use the phet simulation below to answer the questions in your lab report

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#### pdf answer key build an atom elle richard - Oct 09 2023

web the charge of an atom is determined by the ratio of protons to electrons build an atom 1 f 6 play with the simulation to discover what affects the mass number of your atom or ion click on the green sign next to mass to reveal the balance then continue to play the same way you were before

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web mechanical properties of astm a234 wpb pipe fittings tensile strength 60 000 pis 415 mpa yield strength 35 000 psi 240 mpa pressures ratings 150 lbs 300 lbs 600 lbs 900 lbs 1500 lbs 2000 lbs 3000 lbs to 9000 lbs for other grades in astm a234 astm a234 includes seamless and welded pipe fittings

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web mechanical properties elongation a wpb and wpc fittings manufactured from plate shall have a minimum elongation of 17 b for each 1 32 in 0 79 mm decrease in wall thickness below 5 16 in 7 94 mm a deduction of 1 5 for longitudinal and 1 0 for transverse from the values shown above is permitted download the full product catalog

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web mechanical properties of astm a 234 wpb wpc 90 elbows l r 4 sch40 a 234 wpb plywood pallet packaged 1 wpb and wpc pipe fittings manufactured from plates shall have a minimum elongation of 17 2 unless required hardness value doesn t need to be reported 3 manufacture straight tees astm a 234 wpb black painted

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web astm a234 wpb pipe fittings mechanical properties chiefly used in applications involving medium to high temperature pipeline systems the astm a234 standard steel pipe fittings are made from materials including different carbon and alloy steel grades bw a234 wpb fittings chemical composition the sum of cu ni cr and mo shall not exceed astm a234 wpb elbow quality manufacturer of industrial pipe - Mar 30 2022

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#### difference between astm a234 wpb and wpc linkedin - Dec 27 2021

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web specifications astm wrought butt weld fittings a234 a234m grade wpb astm a234 a234m grade wpb chemical properties mechanical properties tensile strength min ksi mpa yield strength min ksi mpa 0 2 offset note a wpb and wpc fittings manufactured from plate shall have a minimum elongation of 17

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