

Name: _____ Period: _____ Date: _____

Student Exploration: Boyle's Law and Charles' Law

Essential Question: How do Boyle's and Charles' laws describe gas pressure?

Exit Ticket: Boyle's Law and Charles' Law (10 minutes)

1. Boyle's Law states that the pressure of a gas is inversely proportional to its volume. If the volume of a gas is doubled, what happens to its pressure?

2. Charles' Law states that the volume of a gas is directly proportional to its temperature. If the temperature of a gas is doubled, what happens to its volume?

Boyle's Law

Boyle's Law states that the pressure of a gas is inversely proportional to its volume. If the volume of a gas is doubled, what happens to its pressure?

1. Boyle's Law states that the pressure of a gas is inversely proportional to its volume. If the volume of a gas is doubled, what happens to its pressure?

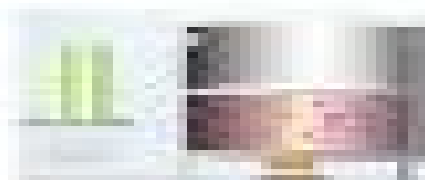
Answer: _____

2. Boyle's Law states that the pressure of a gas is inversely proportional to its volume. If the volume of a gas is doubled, what happens to its pressure?

Boyle's Law states that the pressure of a gas is inversely proportional to its volume. If the volume of a gas is doubled, what happens to its pressure?

3. Boyle's Law states that the pressure of a gas is inversely proportional to its volume. If the volume of a gas is doubled, what happens to its pressure?

4. Boyle's Law states that the pressure of a gas is inversely proportional to its volume. If the volume of a gas is doubled, what happens to its pressure?



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