# **ਜ਼ਗ਼ਗ਼ਗ਼ਗ਼ਜ਼ਗ਼ਜ਼ਗ਼ਜ਼ਜ਼**ਗ਼

Introduction: All compounds can be classified in 2 broad categories:

- 1.) Organic compounds- Contain carbon and hydrogen atoms
- 2.) Inorganic compounds- Can have one or the other, but do not contain both carbon and hydrogen atoms
- A. Most of your body's molecules are organic compounds.
  - Macromolecules are built from small organic compounds the same way a railroad train is built, by linking a lot of smaller units together into long chains.
    - Large carbon compounds are built up from smaller simpler molecules called monomers (mono = one)
    - Monomers can bind to one another to form complex molecules known as polymers (poly = many)
    - A polymer consists of repeated, linked units, which can also bind forming large polymers called Macromolecules. (macro = large )
  - Monomers link to form polymers through a chemical reaction called condensation reaction or dehydration synthesis. During the formation of polymers, Water (H<sub>2</sub>O), is <u>released</u> or is by-product of the reaction.



- The breakdown of some complex molecules, such as polymers, occurs through a process known as hydrolysis.
  - Hydrolysis is the reverse of a condensation reaction. The <u>addition</u> of water, to some polymers
    can break the bonds that hold them together.

#### There are four main types of macromolecules found in living organisms:

- 1.) Carbohydrates
- 2.) Lipids
- 3.) Proteins
- 4.) Nucleic Acids



Name: KEY

#### Carbohydrates

- Composed of carbon, hydrogen, and oxygen atoms in the proportion of 1:2:1
  - General formula: (CH<sub>2</sub>O)<sub>n</sub> where n is the number of carbon atoms.
    - Example: The sugar glucose is a small carbohydrate; its n equals 6. Therefore its chemical formula is C<sub>e</sub>H<sub>±2</sub>O<sub>e</sub>.
  - ii. The building blocks (or monomers) of carbohydrates are monosaccharides.
  - Monosaccharides are simple sugars (saccharide = sugar). Examples:
    - 1. Glucose: commonly found in blood of animals
    - 2. Galactose: a simple sugar found in milk
    - 3. Fructose: commonly found in fruit
      - a. Glucose and Fructose both have the formula C<sub>4</sub>H<sub>12</sub>O<sub>6</sub>, Sometimes compounds may have the same formula, however they have different structures/ arrangements. In such cases, those compounds are called isomers.

# **Biology The History Of Life Answer Key**

**Thomas Flatt, Andreas Heyland** 

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