

2 - Bonding & Molecular Structures

10 20 30 40 50 60 70 80 90 100

Table 4.4 - Some Average Single and Multiple Bond Strengths (kJ/mol)

	H	C	N	O	F	Cl	Br	I	S	Se	Te
H	436										
C		413	347	336	485						
N			391	305	460						
O				443	425						
F					485						
Cl						328					
Br							276				
I								240			
S									265		
Se										226	
Te											226

Multiple Bonds

C=C	614	C≡C	839
C=O	799	C≡N	891
C≡N	891	N≡N	945
O=O	498		
O=O and H ₂ O	500		

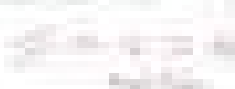
Table 4.5 - Standard Enthalpies of Formation (kJ/mol)

Compound	Standard Enthalpy of Formation	Compound	Standard Enthalpy of Formation
H ₂ O(l)	-285.8	H ₂ O(g)	-241.8
H ₂ O(s)	-291.8		

1. Write the balanced chemical equation for the complete combustion of ethane (C₂H₆).



2. Draw structural formula for each molecule and label the geometry.



3. Calculate the standard enthalpy of formation for the products in the reaction:
 Calculate the standard enthalpy of formation for the products in the reaction.



4. What is the standard enthalpy of formation for CO₂ (g)?

5. Use the bond enthalpies given below to calculate the standard enthalpy of formation for CO₂ (g).

Bonding And Molecular Structure Worksheet Answers South Pasadena

Harry B. Gray



Bonding And Molecular Structure Worksheet Answers South Pasadena:

Chemical Bonds Harry B. Gray, 1994-12-05 This profusely illustrated book by a world renowned chemist and award winning chemistry teacher provides science students with an introduction to atomic and molecular structure and bonding This is a reprint of a book first published by Benjamin Cummings 1973 Structure and Bonding Jack Barrett, 2001 This book explains in non mathematical terms where possible the factors that govern covalent bond formation the lengths and strengths of bonds and molecular shapes Chemical Structure and Bonding Roger L. DeKock, Harry B. Gray, 1989 Designed for use in inorganic physical and quantum chemistry courses this textbook includes numerous questions and problems at the end of each chapter and an Appendix with answers to most of the problems *Chemical Bonding and the Geometry of Molecules* George E. Ryschkewitsch, 1965 **Chemical Bonding and the Geometry of Molecules** George Eugene Ryschkewitsch, 1965 *The Shape and Structure of Molecules* Charles Alfred Coulson, 1982 **Bonding and Molecular Structure** Russell S. Drago, 1971 *A Pictorial Approach to Molecular Bonding* John G. Verkade, 2012-12-06 With the development of accurate molecular calculations in recent years useful predictions of molecular electronic properties are currently being made It is therefore becoming increasingly important for the non theoretically oriented chemist to appreciate the underlying principles governing molecular orbital formation and to distinguish them from the quantitative details as associated with particular molecules It seems highly desirable then that the non theoretician be able to deduce results of general validity without esoteric mathematics In this context pictorial reasoning is particularly useful Such an approach is virtually indispensable if bonding concepts are to be taught to chemistry students early in their careers Undergraduate chemistry majors typically find it difficult to formulate molecular orbital schemes especially delocalized ones for molecules more complicated than diatomics The major reason for this regrettable situation is the general impracticability of teaching group theory before students take organic and inorganic courses wherein the applications of these concepts are most beneficial Consequently many students graduate with the misconception that the ground rules governing bonding in molecules such as NH_3 are somehow different from those which apply to aromatic systems such as C_6H_6 Conversely seniors and many graduate students are usually only vaguely if at all aware that sigma bonding like extended pi bonding can profitably be described in a delocalized manner when discussing the UV photoelectron spectrum of CH_6 for example *The Nature of the Chemical Bond and the Structure of Molecules and Crystals* Linus Pauling, 1963 The Theory of Molecular Structure and Bonding, 1980 Molecular Structure and Conformation I. G. Csizmadia, 1982 Good No Highlights No Markup all pages are intact Slight Shelfwear may have the corners slightly dented may have slight color changes slightly damaged spine **Nature of the Chemical Bond and the Structure of Molecules and Crystals** Linus Pauling, 1967 *The Determination of Molecular Structure* P.J. Wheatley, 1981 **Structure of Molecules and the Chemical Bond** I. A. Kivovich Syrkin, M. E. Diatkina, M. E. Diatkina, 1950 The Determination of Molecular Structure Peter Jaffrey

Wheatley,1962 *MOLECULAR STRUCTURE AND BONDING; THE QUALITATIVE MOLECULAR ORBITAL APPROACH.*
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