

Ans: a) Gold sol

7. What is the unit of rate constant for zero order reaction?

- a) $\text{Mol L}^{-1}\text{s}^{-1}$ b) $\text{Mol}^{-1}\text{L s}^{-1}$ c) s^{-1} d) $\text{Mol}^{-1} \text{L}^{-1}\text{s}^{-1}$ (1)

Ans: $\text{Mol L}^{-1}\text{s}^{-1}$

8. "The limiting molar conductivity of an electrolyte is obtained by adding the limiting molar conductivities of cations and anions of the electrolyte". Name the law. (1)

Ans: a) Kohlrausch's law

9. Calculate the no. of tetrahedral voids, if Mg crystal contains 'n' number of atoms. (1)

Ans: No. of tetrahedral voids = $2n$

Part I SECTION B

(Answer any Four questions from 10 to 13, each question carries 1 scores – non focus area)

10. Which of these is temperature dependent?

- a) Molarity b) Molality c) Mole fraction d) ppm (1)

Ans: a) Molarity

11. What is the basicity of H_3PO_3 ?

- a) 1 b) 2 c) 3 d) 4 (1)

Ans: b) 2

12. $\text{CH}_3\text{-CH}_2\text{-CH}_2\text{-Br} \xrightarrow{\text{alc. KOH}}$? (1)

Ans: $\text{CH}_3\text{-CH=CH}_2$ {Propene}

13. $\text{-----} + \text{CH}_3\text{MgBr} \longrightarrow \text{CH}_3\text{-CH}_2\text{-OMgBr} \xrightarrow{\text{H}_2\text{O}} \text{CH}_3\text{-CH}_2\text{-OH} + \text{MgBr(OH)}$ (1)

Ans: HCHO

Part II SECTION A

(Answer any Two questions from 14 to 17, each question carries 2 scores - focus area)

Bonar Law Chapter 11 Chemistry Answerkey

Julia Schneider



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