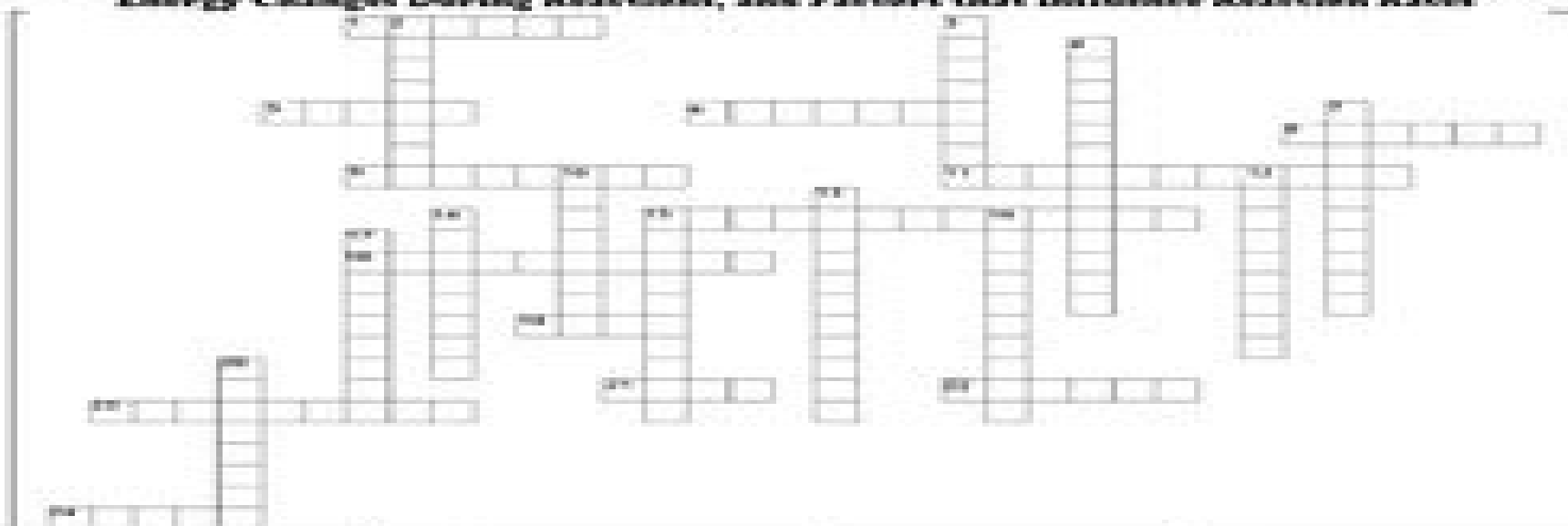


Name: _____

Chemical Reactions Crossword Puzzle Part 2: Four Types of Chemical Reactions, Energy Changes During Reactions, and Factors that Influence Reaction Rates



Across

1. A _____ replacement reaction is when one element replaces another element in a compound.
5. There are _____ factors that influence chemical reaction rates: surface area, temperature and concentration/pressure.
6. The reaction rate will occur slower or faster depending on how much outside _____ area of the reactants are exposed.
8. The Law of Conservation of _____ states that energy cannot be created or destroyed, it can only change its form.
9. The _____ rate is the speed at which a chemical reaction will take place.
11. _____ reactions are chemical reactions that absorb energy from the environment as the reaction is taking place.
13. When the _____ of the reactants is increased, the reaction will happen faster because of the increased amount of solute in the solution.
18. _____ reactions are chemical reactions that release thermal energy as the reaction is taking place.
19. Exothermic reactions require _____ activation energy to start the chemical reaction.
21. Endothermic reactions require _____ activation energy to start the chemical reaction.
22. A _____ replacement reaction is when elements in two compounds will switch places forming two new compounds.
23. A _____ reaction is when two or more substances combine to form one compound.
24. The released energy in a combustion reaction is usually in the form of _____ energy or thermal energy.

Down

2. An _____ in temperature means collisions between particles will happen more often, and the chemical reaction will occur faster.
3. An endothermic reaction absorbs energy which results in a temperature _____ from the surrounding environment.
4. A _____ reaction is when one compound breaks down and forms two or more substances.
7. Substances that slow or stop a chemical reaction by increasing the activation energy of the reaction.
10. An exothermic reaction releases energy which results in a temperature _____ from the surrounding environment.
12. Chemical _____ occur when substances are changed into new substances.
13. It is a measure of the average kinetic energy of the particles that make up a substance.
14. The ending substances in a chemical reaction.
15. A _____ reaction is a reaction in which a substance combines with oxygen and releases energy.
16. _____ energy is the minimum amount of energy needed to start a chemical reaction.
17. The starting substances in a chemical reaction.
20. A substance that increases the reaction rate of a reaction by lowering the activation energy of the reaction.

Crossword Chemical Reactions Science Spectrum 2008

Answer

J Ma



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web 7 x borland turbo pascal is a pascal programming language compiler with an integrated development environment targeted at the hobbyist and entry level market there were versions for cp m and dos for a time it was sold along side their professional borland pascal product line