

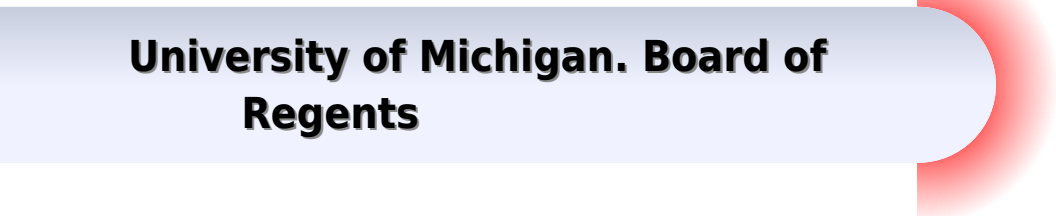
Cp physics - Spring Final Review (second semester topics)**Multiple Choice**

Identify the choice that best completes the statement or answers the question.

- _____ 1. Which of the following is a direct cause of a substance's temperature increase?
- Energy is removed from the particles of the substance.
 - Kinetic energy is added to the particles of the substance.
 - The number of atoms and molecules in a substance changes.
 - The volume of the substance decreases.
- _____ 2. What happens to the internal energy of an ideal gas when it is heated from 0°C to 4°C ?
- It increases.
 - It decreases.
 - It remains constant.
 - It is impossible to determine.
- _____ 3. Which of the following is proportional to the kinetic energy of atoms and molecules?
- elastic energy
 - temperature
 - potential energy
 - thermal equilibrium
- _____ 4. As the temperature of a substance increases, its volume tends to increase due to
- thermal equilibrium.
 - thermal expansion.
 - thermal contraction.
 - thermal contraction.
- _____ 5. What is the temperature of a system in thermal equilibrium with another system made up of water and steam at 1 atm of pressure?
- 0°F
 - 273 K
 - 0 K
 - 100°C
- _____ 6. Energy transferred as heat occurs between two bodies in thermal contact when they differ in which of the following properties?
- mass
 - specific heat
 - density
 - temperature
- _____ 7. Which of the following terms describes a transfer of energy?
- heat
 - internal energy
 - temperature
 - kinetic energy
- _____ 8. The use of fiberglass insulation in the outer walls of a building is intended to minimize heat transfer through what process?
- conduction
 - radiation
 - convection
 - vaporization
- _____ 9. How is energy transferred as heat always directed?
- from an object at low temperature to an object at high temperature
 - from an object at high temperature to an object at low temperature
 - from an object at low kinetic energy to an object at high kinetic energy
 - from an object with higher mass to an object of lower mass
- _____ 10. What three properties of a substance affect the amount of energy transferred as heat to or from the substance?
- volume, temperature change, specific heat capacity
 - density, temperature change, specific heat capacity
 - mass, temperature change, specific heat capacity
 - mass, temperature change, latent heat

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Chemistry for Degree Students B.Sc. Semester - II (As per CBCS) Madan R.L., 2017 This textbook has been designed to meet the needs of B Sc Second Semester students of Chemistry as per the UGC Choice Based Credit System CBCS With its traditional approach to the subject this textbook lucidly explains principles of chemistry Important topics such as chemical energetics chemical ionic equilibrium aromatic hydrocarbons alkyl aryl halides alcohols phenols ethers aldehydes and ketones are aptly discussed to give an overview of physical and organic chemistry Laboratory work has also been included to help students achieve solid conceptual understanding and learn experimental procedures **Essential Advanced General**

Mathematics Third Edition Enhanced TIN/CP Version Michael Evans, Kay Lipson, Douglas Wallace, Sue Avery, 2011-04 The Essential VCE Mathematics series has a reputation for mathematical excellence with an approach developed over many years by a highly regarded author team of practising teachers and mathematicians This approach encourages understanding through a wealth of examples and exercises with an emphasis on VCE examination style questions New in the enhanced versions TI Nspire OS3 and Casio ClassPad calculator explanations examples and problems are integrated into the text Page numbers in the printed text reflect the previous TI nspire and Casio ClassPad version allowing for continuity and compatibility Digital versions of the student text are available in Interactive HTML and PDF formats through Cambridge GO

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The Johns Hopkins University Circular, 1946 Includes University catalogues President's report Financial report registers announcement material etc **College Credit Recommendations**, 2002 *General Register* University of Michigan, 1946 Announcements for the following year included in some vols *Catalogue of the University of Michigan* University of Michigan, 1955 Announcements for the following year included in some vols *AEC Authorizing Legislation FY62* United States. Congress. Joint Committee on Atomic Energy. Subcommittee on Legislation, 1961 Considers 87 H R 6744 87 H R 7576 87 S 2043 87 S 1774 **University of Michigan Official Publication**, 1951 **Correspondence Courses**

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Johns Hopkins University Circulars Johns Hopkins University,1945 **Proceedings of the Board of Regents** University of Michigan. Board of Regents,1957 **Hearings** United States. Congress. Joint Committee ...,1961 **AEC Authorizing Legislation** United States. Congress. Joint Committee on Atomic Energy,1961 **Hearings and Reports on Atomic Energy** United States. Congress. Joint Committee on Atomic Energy,1946 [Resources in Education](#) ,1979

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