

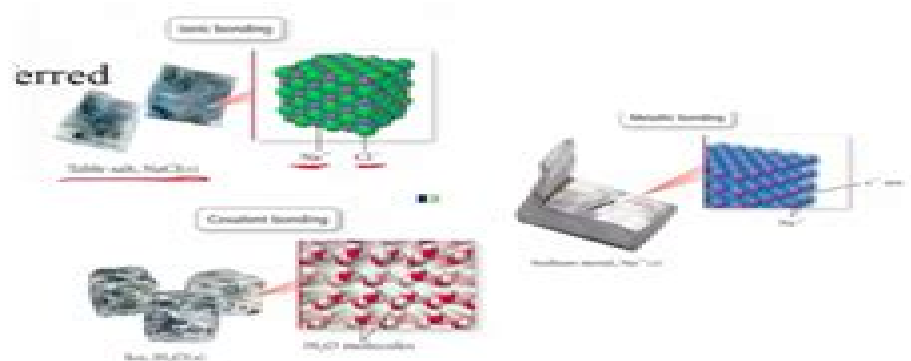
## Chemical Bonding 1: Lewis Theory Introduction to Bonding

### Types of Chemical Bonding

- **Three Theories for Bonding :**
  - 1) Lewis Theory
  - 2) Valence Bond Theory
  - 3) Molecular Orbital Theory

### Summary of the Types

- **Ionic Bonding**
  - Metal to nonmetal
  - Electrons are transferred
- **Covalent Bonding**
  - Nonmetal to nonmetal
  - Electrons are shared
- **Metallic Bonding**
  - Metal to metal



### Lewis Dot Symbols

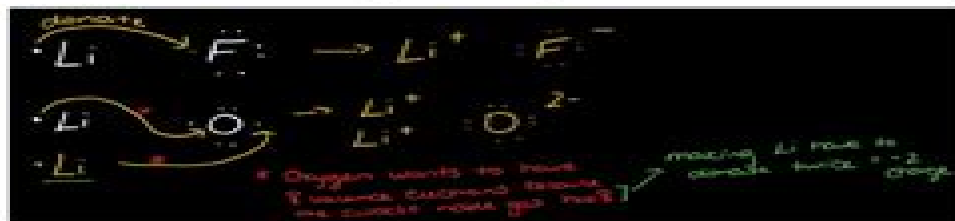
- Symbols plus one dot for each valence electron

1 1A	2 2A	3 3A	4 4A	5 5A	6 6A	7 7A	8 8A	9 9A	10 10A	11 1B	12 2B	13 3A	14 4A	15 5A	16 6A	17 7A	18 8A
H	He											B	C	N	O	F	Ne
Li	Be											Al	Si	P	S	Cl	Ar
Na	Mg											Ga	Ge	As	Se	Br	Kr
K	Ca											In	Sn	Sb	Te	I	Xe
Rb	Sr											Tl	Pb	Bi	Po	At	Rn
Cs	Ba																
Fr	Ra																

### Ionic Bonding Using Lewis Theory

#### Ionic Bonding: Lewis Structures and Lattice Energies

- **Ionic Bond:** the electrostatic force that holds ions together in an ionic compound
- We can use Lewis Dot symbols to represent an ionic bond being formed



# Core Teaching Resources Chemistry Chapter 7

**Patrick Fullick**



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