

How does bonding affect behavior?

1. Metals (Metallic Bond)

The clustering of valence electrons into metals

2. How valence electrons interact in a metal crystal lattice

(2 part bond: 1 Strong, 2, 3 weaker, 4, 5, 6, 7, 8, 9, 10)

3. Crystal's interatomic forces: metallic bond and ionic bond

Crystal's structure is determined by the balance of attractive and repulsive forces. (Metals: 1, 2, 3, 4, 5, 6, 7, 8, 9, 10)

Crystal's structure is determined by the balance of attractive and repulsive forces. (Metals: 1, 2, 3, 4, 5, 6, 7, 8, 9, 10)

4. Crystal's structure: metals, alloys, and ionic crystals

(Single bond: 1, 2, 3, 4, 5, 6, 7, 8, 9, 10) (Compound: 1, 2, 3, 4, 5, 6, 7, 8, 9, 10)

Double bond: 1, 2, 3, 4, 5, 6, 7, 8, 9, 10 (Compound: 1, 2, 3, 4, 5, 6, 7, 8, 9, 10)

Triple bond: 1, 2, 3, 4, 5, 6, 7, 8, 9, 10 (Compound: 1, 2, 3, 4, 5, 6, 7, 8, 9, 10)

5. Crystal's structure and properties: metals, alloys, and ionic crystals

Crystal's structure and properties: metals, alloys, and ionic crystals

Crystal's structure and properties: metals, alloys, and ionic crystals



6. Crystal's structure and properties: metals, alloys, and ionic crystals

Crystal's structure and properties: metals, alloys, and ionic crystals



7. Crystal's structure and properties: metals, alloys, and ionic crystals



8. Crystal's structure and properties: metals, alloys, and ionic crystals

Crystal's structure and properties: metals, alloys, and ionic crystals

9. Crystal's structure and properties: metals, alloys, and ionic crystals

Crystal's structure and properties: metals, alloys, and ionic crystals

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