Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

There are eight markers in a full set, but Flora and Frank each only have seven markers.
 Flora is missing the red marker, and Frank is missing the blue marker.

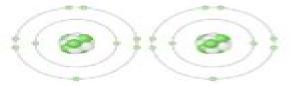
What can they do so that each has a full set of markers? If frank shares his red marker with Flora then she can she her blue marker with Frank

2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set? Olivia shares her purple and green markers with Otta and then Otta can share his brown and black markers with OLivia

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the Covalent Bonds Gizmo, atoms form bonds in this way.

To begin, check that Fluorine is selected from the Select a substance menu. Click Play () to see the electrons orbiting the nucleus of each atom.



- The outermost electrons in each atom are called valence electrons. How many valence electrons does each fluorine atom have? The fluorine atoms have 7 valence electrons
- Click Pause (*). Drag a valence electron from the left atom to the right atom. Click Play.

What happens? Electrons were dragged now orbit both of the atoms

Click Pause, drag a valence electron from the right atom to the left, and then click Play.

What happens now? The pair of electrons will now orbit both of the atoms



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Patrick Vollmar

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Table of Contents Covalent Bonds Gizmo Activity B Gizmo

- 1. Understanding the eBook Covalent Bonds Gizmo Activity B Gizmo
 - The Rise of Digital Reading Covalent Bonds Gizmo Activity B Gizmo
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Covalent Bonds Gizmo Activity B Gizmo
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Covalent Bonds Gizmo Activity B Gizmo
 - User-Friendly Interface

- 4. Exploring eBook Recommendations from Covalent Bonds Gizmo Activity B Gizmo
 - Personalized Recommendations
 - Covalent Bonds Gizmo Activity B Gizmo User Reviews and Ratings
 - o Covalent Bonds Gizmo Activity B Gizmo and Bestseller Lists
- 5. Accessing Covalent Bonds Gizmo Activity B Gizmo Free and Paid eBooks
 - Covalent Bonds Gizmo Activity B Gizmo Public Domain eBooks
 - Covalent Bonds Gizmo Activity B Gizmo eBook Subscription Services
 - o Covalent Bonds Gizmo Activity B Gizmo Budget-Friendly Options
- 6. Navigating Covalent Bonds Gizmo Activity B Gizmo eBook Formats
 - o ePub, PDF, MOBI, and More
 - Covalent Bonds Gizmo Activity B Gizmo Compatibility with Devices
 - Covalent Bonds Gizmo Activity B Gizmo Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Covalent Bonds Gizmo Activity B Gizmo
 - Highlighting and Note-Taking Covalent Bonds Gizmo Activity B Gizmo
 - Interactive Elements Covalent Bonds Gizmo Activity B Gizmo
- 8. Staying Engaged with Covalent Bonds Gizmo Activity B Gizmo
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Covalent Bonds Gizmo Activity B Gizmo
- 9. Balancing eBooks and Physical Books Covalent Bonds Gizmo Activity B Gizmo
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Covalent Bonds Gizmo Activity B Gizmo
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Covalent Bonds Gizmo Activity B Gizmo
 - Setting Reading Goals Covalent Bonds Gizmo Activity B Gizmo
 - Carving Out Dedicated Reading Time

- 12. Sourcing Reliable Information of Covalent Bonds Gizmo Activity B Gizmo
 - Fact-Checking eBook Content of Covalent Bonds Gizmo Activity B Gizmo
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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