



Name: \_\_\_\_\_

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## Student Exploration: Covalent Bonds

**Vocabulary:** covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? If Frank shares his red marker with Flora then she can share her blue marker with Frank.


2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set? Olivia shares her purple and green markers with Otto and then Otto can share his brown and black markers with Olivia.

### Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** () to see the electrons orbiting the nucleus of each atom.



1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have? The fluorine atoms have 7 valence electrons.
2. Click **Pause** (). Drag a valence electron from the left atom to the right atom. Click **Play**. What happens? Electrons were dragged now orbit both of the atoms.
3. Click **Pause**, drag a valence electron from the right atom to the left, and then click **Play**. What happens now? The pair of electrons will now orbit both of the atoms.



# Covalent Bonds Gizmo Activity B Gizmo

**Patrick Vollmar**



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