

## Coulombic Attraction

What variables will affect the force of attraction between charged particles?

### Why?

**Coulombic attraction** is the attraction between oppositely charged particles. For example, the protons in the nucleus of an atom have attraction for the electrons surrounding the nucleus. This is because the protons are positive and the electrons are negative. The attractive force can be weak or strong. In this activity, you will explore the strength of attraction between protons and electrons in various atomic structures.

### Model 1 – Distance and Attractive Force

		Force of Attraction (Newtons)
A	 0.10 nm	$2.30 \times 10^{-8}$
B	 0.20 nm	$0.58 \times 10^{-8}$
C	 0.30 nm	$0.26 \times 10^{-8}$

1. What subatomic particles do these symbols represent in Model 1?

 protons       electrons

2. Would you expect to observe attraction or repulsion between the subatomic particles in Model 1?

Attraction.



3. Consider the data in Model 1.

- a. What are the independent and dependent variables in the data?

Distance      Force of Attraction

- b. Write a complete sentence that describes the observed relationship between the independent and dependent variables in Model 1.

As distance increases, force of attraction decreases (inverse)

4. If the distance between a proton and electron is 0.50 nm, would you expect the force of attraction to be greater than or less than  $0.26 \times 10^{-8}$  N?

less than  $0.26 \times 10^{-8}$  N

5. If two protons are 0.10 nm away from one electron, would you expect the force of attraction to be greater than or less than  $2.30 \times 10^{-8}$  N?

Greater than  $2.30 \times 10^{-8}$  N



# Coulombic Attraction Answers

**C. A. Trapp, M. P. Cady, Carmen Giunta**

## **Coulombic Attraction Answers:**

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information presented is supported through citations to chemical literature Problems are incorporated throughout the text and full worked out solutions are presented for every problem International Union of Pure and Applied Chemistry style and technical guidelines are followed throughout the text The problems text and presentation are based on years of classroom refinement of teaching pedagogy This textbook is aimed at an advanced high school or general college audience aiming to engage students more directly in the work of chemistry William Tucker s passion for chemistry was inspired by his high school teacher Gary Osborn He left Maine to pursue Chemistry at Middlebury College and after graduating in 2010 he decided to pursue a PhD in Organic Chemistry at the University of Wisconsin Madison At the University of Wisconsin Madison he worked in the laboratory of Dr Sandro Mecozzi where he developed semifluorinated triphilic surfactants for hydrophobic drug delivery After earning his PhD in 2015 he took a fellowship at Boston University as a Postdoctoral Faculty Fellow There he co taught organic chemistry while working in the laboratory of Dr John Caradonna In the Caradonna l boratory he worked on developing a surface immobilized iron oxidation catalyst for the oxidation of C H bonds using dioxygen from the air as the terminal oxidant Throughout all of this work his passion has always been for teaching and working with students both in and out of the classroom He has been lucky for the past six years to work at Concord Academy where his students have through their questions pushed him to think deeper and more critically about chemistry Their curiosity inspires him and their inquisitiveness inspired his writing

**Chemistry and Physics for Nurse Anesthesia, Second Edition** David Shubert, PhD, John Leyba, PhD, 2013-03-15 Print CourseSmart

**Student's Solutions Manual to Accompany Atkins' Physical Chemistry** C. A. Trapp, M. P. Cady, Carmen Giunta, 2010 This solutions manual provides the authors detailed solutions to exercises and problems in physical chemistry It comprises solutions to exercises at the end of each chapter and solutions to numerical theoretical and additional problems

**Get Through Final FRCA: MCQs** Nawal Bahal, Mubeen Khan, Alik Manoras, 2010-06-25 With more than 950 questions and containing diagrams found in the actual exam this book encompasses the syllabus of the Final FRCA as set by the Royal College of Anaesthetists including five complete examinations each comprising of 90 MCQs Questions have been modelled on actual exam questions so that they accurately reflect the style and level

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Scheiner, 2015-04-16 Computational methods and in particular quantum chemistry have taken the lead in our growing understanding of noncovalent forces as well as in their categorization This volume describes the current state of the art in terms of what we now know and the current questions requiring answers in the future Topics range from very strong ionic to very weak CH interactions In the intermediate regime forces to be considered are H bonds particularly CH O and OH metal halogen chalcogen pnictogen and tellurium bonds aromatic stacking dihydrogen bonds and those involving radicals Applications include drug development and predictions of crystal structure **Plasma-Surface Interactions and Processing of**

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