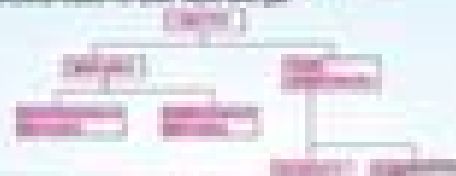


1. CLASSIFICATION OF MATTER

- Matter can exist in three physical states: solid, liquid and gas.



- Various sub-classifications of matter are given below:
- Matter is classified into pure and impure.
- Pure substances are further classified into elements and compounds.
- Compounds are further classified into ionic and covalent.
- Compounds are further classified into simple and complex.
- Compounds are further classified into organic and inorganic.
- Compounds are further classified into natural and synthetic.

2. PROPERTIES OF MATTER AND THEIR MEASUREMENT

- Properties of matter are physical and chemical properties.
- Physical properties are those which can be measured without changing the chemical composition of the substance.
- Chemical properties are those which can be measured by changing the chemical composition of the substance.
- Some physical properties are: mass, volume, density, etc.
- Some chemical properties are: reactivity, etc.
- Quantitative measurement of physical properties is called measurement of matter.
- Measurement of matter is called measurement.
- Measurement of matter is called measurement.

UNIT 1: PHYSICAL PROPERTIES OF MATTER

Mass (kg)	Volume (m ³)	Density (kg/m ³)	Temperature (°C)	Pressure (Pa)
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UNIT 2: CHEMICAL PROPERTIES

- 1. $2H_2 + O_2 \rightarrow 2H_2O$ (2 molecules of hydrogen react with 1 molecule of oxygen to form 2 molecules of water)

3. SIGNIFICANT FIGURES

- Significant figures are those digits which are known with certainty plus one additional uncertain or doubtful digit.
- Rules for determining the number of significant figures:
 - Non-zero digits are significant.
 - Zeros between non-zero digits are significant.
 - Zeros at the end or right of a number are significant if they are on the right side of a decimal point.
 - Leading zeros are not significant.
 - In a number written in scientific notation, all digits are significant.
 - Exact values are the numbers of units or measurements for which there is no uncertainty.
 - Round-off to the nearest value of a significant figure to the first value of the next digit.
 - In addition and subtraction of significant figures, the result cannot have more digits to the right of the decimal point than either of the original numbers.
 - In multiplication and division of significant figures, the result must be rounded off to the same number of significant figures as the least number of significant figures in the original numbers.

Aakash Chemistry Lab Manual

**Mark J. Mezger, Kay J. Tindle, Michelle
Pantoya, Lori J. Groven, Dilhan Kalyon**

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Reviewing **Aakash Chemistry Lab Manual**: Unlocking the Spellbinding Force of Linguistics

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