Using means, scientists have been able to construct different models of the The atomic model has taken year years to build and people still tweak it now and again, hopefully making it more accused 1802 - At the beginning of the 19 <sup>th</sup> century, Englishman built an atomic model called the " is made of only the strength of the
1802 - At the beginning of the 19 <sup>th</sup> century, Englishman built an atomic model called the "
1802 - At the beginning of the 19 <sup>th</sup> century, Englishman  built an atomic model called the "
built an atomic model called the "
According to his theory, each type of is made of only is
According to his theory, each type of is made of only
type of Also each type of atom could be put
together into a group called an and every atom of a
type of Also each type of atom could be put together into a group called an and every atom of a element was For example, gold atoms mak, and so on. Dalton also
and iron atoms make, and so on. Dalton als
believed that atoms could not be, divided into sma
parts, or
1897 - About 100 years after Dalton, British scientists
discovered the Thompson stated that negatively
discovered the Thompson stated that negatively charged electrons must be stuck all throughout a positively charged area. Think of
the atom is cookie dough and the are chocolate chip
The discovery of electrons was a big deal. But it left some bi questions- what were
positive particles holding those electrons in place?
1909 - In 1909, a New Zealand born British scientist named
and his team set out to answer some of these question
After performing in depth experiments using gold foil, Rutherford determined tha
gold atom must have some very small charged mass. I
named this mass the He called the positively charge
nucleus particles, and said that the
were scattered in a
the nucleus.
James Chadwick, a student of Rutherford's, found new particles in the nucleus tha
weren't affected by an electric field at all. Chadwick called these uncharged partic
Also in the early 20th century, scientists discovered that electrons are arranged in
energy levels. The lower energy levels are closer to the
can hold just a few The higher energy levels are
farther from the nucleus and can hold more
Electrons are so small and that their energy levels are i
neat planet-like orbits around the nucleus. Instead, they exist in a region called the
electron cloud. The atomic model is constantly evolving as scientists discover more

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