Date: Name:

Student Exploration: Boyle's Law and Charles' Law

Vocabulary: absolute zero. Boyle's law. Charles' law. Gay-Lussac's law. Kelvin scale, pressure

Prior Knowledge Question (Do this BEFORE using the Gizmo.)

A small helium tank measures about two feet (60 cm) high. Yet it can fill over 50 balloons! How can such a small tank contain enough helium to fill so many balloons?

The balloons don't need a lot of helium, a small container can hold a lot of helium.

Gizmo Warm-up

The Boyle's Law and Charles' Law Gizmo™ shows a container of gas. In the container, the small purple spheres represent molecules.



speed? No, some move faster and others move slower



How do the particles interact with the walls and lid of the container? They bounce off the. walls and lid of the container.

These interactions contribute to the pressure on the walls of the container. Pressure is defined as force per unit area. The SI units of pressure are newtons per square meter (N/m²), or pascals (Pa).

- Slowly drag the temperature (T) slider back and forth. (Note: In this Gizmo, the Kelvin scale is used to measure temperature. On the Kelvin scale, 0 degrees is absolute zero, the coldest possible temperature. Absolute zero is equal to -273.15 °C or -459.67 °F)
 - A. How does the change in temperature affect the speed of the molecules? Higher temperature, faster speed, Lower temperature, slower speed,
 - B. How does the change in temperature affect the volume of the container? <u>Higher</u>. temperature, higher volume. Lower temperature, less volume.

Boyle S Law Gizmo Answer Key

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Boyle S Law Gizmo Answer Key:

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