

air.

The Respiratory System

oxygen

Use the words in the box to fill in the blanks.

The company of the company of the first transfer of the first	and the state of t	bronchi	pharynx
carbon dioxide hiccup	yawii	water vapor	
mouth	sneeze	blood	nose
modui	SHEEZE	bicou	
All animals need		to make energy fr	om food. We get
this oxygen from the		that we breathe. In order to get	
the oxygen into the	blood where it a	an be transported to	the rest of the
body, the air trave	ls through a syste	em of organs called ti	he
	system.	11.5	
			211-06721
		enters the body thro	
		From there	
the	, which force	s air into the	and
food into the esoph	agus. The air tra	ivels down the trache	a into two
		vels down the trache and then on	
branching tubes ca	lled	and then on	into the
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In the lungs oxyger	lled n from the air en	and then on ters the	into the At the
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In the lungs oxyger same time, the was then leaves the boo	lled trom the air en te gas dy when you also leaves	ters the leav	into the At the res the blood and Some chale, which is
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inhale

exhale

Bogglesworld The Respiratory System Key

Cassie M. Lawton

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buffers typically consist of an acid base pair with the acid and base differing by the presence or absence of a proton a conjugate acid base pair for instance one of the buffers that maintain the ph of human blood involves carbonic acid h 2 co 3 and its conjugate base the bicarbonate ion hco 3

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jul 19 2022 47428 acids and bases are common substances found in many every day items from fruit juices and soft drinks to soap in this chapter we will examine the properties of acids and bases and learn about the chemical nature of these important compounds we will cover ph and how to calculate the ph of a solution

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acids bases and enzymes many acids and bases in living things provide the ph that enzymes need enzymes are biological catalysts that must work effectively for biochemical reactions to occur most enzymes can do their job only at a certain level of acidity cells secrete acids and bases to maintain the proper ph for enzymes to do their work

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acid base titration difference between acid and base acids and bases definition an acid is any hydrogen containing substance that is capable of donating a proton hydrogen ion to another substance a base is a molecule or ion able to accept a hydrogen

ion from an acid acidic substances are usually identified by their sour taste

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a binary acid is an acid that consists of hydrogen and one other element the most common binary acids contain a halogen the acid name begins with the prefix hydro followed by the base name of the anion followed by the suffix ic

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oct 18 2023 most of the acids are solid it has a sour taste they produce hydrogen after reacting with active metals like zinc magnesium aluminium or iron acids have a low ph value i e below 7 salts and water are produced when an acid reacts with base or alkalis most of the acids are solids

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arrhenius s definition of acids and bases the earliest definition of acids and bases is arrhenius s definition which states that an acid is a substance that forms hydrogen ions h when dissolved in water and a base is a substance that forms hydroxide ions oh when dissolved in water

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the reaction of an acid and a base is called a neutralization reaction although acids and bases have their own unique chemistries the acid and base cancel each other s chemistry to produce a rather innocuous substance water in fact the general reaction between an acid and a base is ce acid base water salt nonumber

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