

SAMPLE PAPER (2023 -24)

CHEMISTRY THEORY (043)

Max. Marks:70

Time: 3 hours

General Instructions:

Read the following instructions carefully.

- (a) There are 33 questions in this question paper with internal choice.
 - (b) SECTION A consists of 16 multiple –choice questions carrying 1 mark each.
 - (c) SECTION B consists of 5 short answer questions carrying 2 marks each.
 - (d) SECTION C consists of 7 short answer questions carrying 3 marks each.
 - (e) SECTION D consists of 2 case – based questions carrying 4 marks each.
 - (f) SECTION E consists of 3 long answer questions carrying 5 marks each.
 - (g) All questions are compulsory.
 - (h) Use of log tables and calculators is not allowed.
-

SECTION A

The following questions are multiple –choice questions with one correct answer. Each question carries 1 mark. There is no internal choice in this section.

- 1. Which of the following solutions will have the highest conductivity at 298 K?
 - (a) 0.01 M HCl solution
 - (b) 0.1 M HCl solution
 - (c) 0.01 M CH_3COOH solution
 - (d) 0.1 M CH_3COOH solution



Identify A and B:

- (a) A = 1-phenylethanal, B = acetophenone
 - (b) A = Benzophenone, B = formaldehyde
 - (c) A = Benzaldehyde, B = Acetophenone
 - (d) A = Benzophenone, B = Acetophenone
- 3. The vitamins which can be stored in our body are:
 - (a) Vitamin A, B, D and E
 - (d) Vitamin A, C, D and K
 - (c) Vitamin A, B, C and D
 - (d) Vitamin A, D, E and K
- 4. What is IUPAC name of the ketone A, which undergoes iodoform reaction to give $\text{CH}_3\text{CH}=\text{C}(\text{CH}_3)\text{COONa}$ and yellow precipitate of CHI_3 ?
 - (a) 3-Methylpent-3-en-2-one
 - (b) 3-Methylbut-2-en- one
 - (c) 2, 3-Dimethylethanone
 - (d) 3-Methylpent-4-one
- 5. Which of the following is not correct?
 - (a) In haloarenes, the electron pairs on halogen atom are in conjugation with π -electrons of the ring.
 - (b) The carbon-magnesium bond is covalent and non-polar in nature.
 - (c) During $\text{S}_\text{N}1$ reaction, the carbocation formed in the slow step being sp^2 hybridised is planar.
 - (d) Out of $\text{CH}_2=\text{CH}-\text{Cl}$ and $\text{C}_6\text{H}_5\text{CH}_2\text{Cl}$, $\text{C}_6\text{H}_5\text{CH}_2\text{Cl}$ is more reactive towards $\text{S}_\text{N}1$ reaction

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Table of Contents Xtreem Papers For Olevel Exam

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 - The Rise of Digital Reading Xtreem Papers For Olevel Exam
 - Advantages of eBooks Over Traditional Books
2. Identifying Xtreem Papers For Olevel Exam
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Xtreem Papers For Olevel Exam
 - User-Friendly Interface
4. Exploring eBook Recommendations from Xtreem Papers For Olevel Exam
 - Personalized Recommendations

- Xtreem Papers For Olevel Exam User Reviews and Ratings
- Xtreem Papers For Olevel Exam and Bestseller Lists
- 5. Accessing Xtreem Papers For Olevel Exam Free and Paid eBooks
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 - Xtreem Papers For Olevel Exam Budget-Friendly Options
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 - Xtreem Papers For Olevel Exam Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Xtreem Papers For Olevel Exam
 - Highlighting and Note-Taking Xtreem Papers For Olevel Exam
 - Interactive Elements Xtreem Papers For Olevel Exam
- 8. Staying Engaged with Xtreem Papers For Olevel Exam
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Xtreem Papers For Olevel Exam
- 9. Balancing eBooks and Physical Books Xtreem Papers For Olevel Exam
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Xtreem Papers For Olevel Exam
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Xtreem Papers For Olevel Exam
 - Setting Reading Goals Xtreem Papers For Olevel Exam
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Xtreem Papers For Olevel Exam
 - Fact-Checking eBook Content of Xtreem Papers For Olevel Exam

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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