

# **GRADE 10 PHYSICAL SCIENCES** NOVEMBER EXAMINATION PAPER 2: CHEMISTRY

#### SECTION A: QUESTION 1

Various options are provided as possible arewers to the following questions. Choose the correct answer and write CNLY the LETTER (A-D) next to the QUESTION NUMBER in your answer book.

1.1	Which of the following is an example of a mixture?			
	A	dicibled water		
	0	damond		
	C	sodum-chknide		

- Dement with the electron configuration: 1st, 2st, 2st
  - oxygen

carbon.

- helium.
- sitrages.
- The chemical formula of sodium suffice:
  - Next.
  - No.5
  - No.5
  - No.S.

1.4 Strongest intermole:	sular force found in water?	H
--------------------------	-----------------------------	---

- lon-dipole
- Dipole-dipole
- Van der Waals.
- Hydrogen bond

#### 1.5 The particles of a gas ....

- are Sightly packed.
- vibrate against each other.
- move freely and rarely touch.
- move over one another.

#### 1.6 Element with the highest electronegativity?

- Hydrogen.
- Hebian.
- Oxygen.
- Fluoring

#### 1.7 When an atom of Group 16 ionises to become atom the

- mass number of the atom will increase.
- atomic rumber of the atom will decrease
- number of electrons will decrease
- number of electrons will increase

#### 1.8 An element with the following Authou diagram will be classified as a:

1	ı.		31
1	14	24	20
			21
11			31

Construction .

- noble gas
- halogen.
- alkali sarth-metal
- alkali metali

# **Question Paper 2 November 2014 Physical Science**

Jim Breithaupt, Catherine Jones

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