

## LIMITING REAGENT Practice Problems

1. At high temperatures, sulfur combines with iron to form the brown-black iron (II) sulfide:  

$$\text{Fe (s)} + \text{S (l)} \rightarrow \text{FeS (s)}$$

In one experiment, 7.62 g of Fe are allowed to react with 8.67 g of S.

- What is the limiting reagent, and what is the reactant in excess?
- Calculate the mass of FeS formed.

2. Acrylonitrile,  $\text{C}_3\text{H}_3\text{N}$ , is the starting material for the production of a kind of synthetic fiber (acrylics) and can be made from propylene,  $\text{C}_3\text{H}_6$ , by reaction with nitric oxide, NO, as follows:



What mass of  $\text{C}_3\text{H}_3\text{N}$  can be made when 21.6 g of  $\text{C}_3\text{H}_6$  react with 21.6 g of nitric oxide?

3. Calculate the percent yield for the reaction:  $\text{P}_4 (\text{s}) + 6 \text{Cl}_2 (\text{g}) \rightarrow 4 \text{PCl}_3 (\text{l})$  if 75.0 g of phosphorus reacts with excess chlorine gas to produce 111.0 g of phosphorus trichloride.

4. Formic acid,  $\text{HCHO}_2$ , burns in oxygen to form carbon dioxide and water as follows:



If a 3.15-g sample of formic acid was burned in 2.0 L of oxygen, what volume of carbon dioxide would be produced? (Assume the reaction occurs at standard temperature and pressure, STP.)

5. Zinc metal reacts with hydrochloric acid to produce zinc chloride and hydrogen gas.
- Balance the following reaction:  $\text{Zn (s)} + 2\text{HCl (aq)} \rightarrow \text{ZnCl}_2 (\text{aq}) + \text{H}_2 (\text{g})$
  - A 3.50-g sample of zinc metal is allowed to react with 2.50 g of hydrochloric acid.

Complete the following table:

Reactants/products	Zn (grams)	HCl (grams)	ZnCl <sub>2</sub> (grams)	H <sub>2</sub> (L)
Before reaction	3.5	2.5	0	0
After reaction	1.26 g	1.26	4.68	0.768

6. Consider the reaction:  $\text{MnO}_2 + 4 \text{HCl} \rightarrow \text{MnCl}_2 + \text{Cl}_2 + 2 \text{H}_2\text{O}$   
 If 0.45 mols of  $\text{MnO}_2$  can react with 48.2 g of HCl, how many grams of  $\text{Cl}_2$  could be produced?
7. One of the components of the fuel mixture on the Apollo lunar module involved a reaction with hydrazine,  $\text{N}_2\text{H}_4$ , and dinitrogen tetroxide,  $\text{N}_2\text{O}_4$ . If the balanced equation for this reaction is



What volume of  $\text{N}_2$  gas (measured at STP) would result from the reaction of 1500 kg of hydrazine and 1000 kg of  $\text{N}_2\text{O}_4$ ?



# Limiting Reactant Problems And Answers High School

**Priya Minhas**





## **Limiting Reactant Problems And Answers High School:**

**Multiple Solution Methods for Teaching Science in the Classroom** Stephen DeMeo, 2008 For the first time in science education the subject of multiple solution methods is explored in book form While a multiple method teaching approach is utilized extensively in math education there are very few journal articles and no texts written on this topic in science Teaching multiple methods to science students in order to solve quantitative word problems is important for two reasons First it challenges the practice by teachers that one specific method should be used when solving problems Secondly it calls into question the belief that multiple methods would confuse students and retard their learning Using a case study approach and informed by research conducted by the author this book claims that providing students with a choice of methods as well as requiring additional methods as a way to validate results can be beneficial to student learning A close reading of the literature reveals that time spent on elucidating concepts rather than on algorithmic methodologies is a critical issue when trying to have students solve problems with understanding It is argued that conceptual understanding can be enhanced through the use of multiple methods in an environment where students can compare evaluate and verbally discuss competing methodologies through the facilitation of the instructor This book focuses on two very useful methods proportional reasoning PR and dimensional analysis DA These two methods are important because they can be used to solve a large number of problems in all of the four academic sciences biology chemistry physics and earth science This book concludes with a plan to integrate DA and PR into the academic science curriculum starting in late elementary school through to the introductory college level A challenge is presented to teachers as well as to textbook writers who rely on the single method paradigm to consider an alternative way to teach scientific problem solving **A Concrete Stoichiometry**

**Unit for High School Chemistry** Jennifer Louise Pakkala, 2006 **High School Chemistry Unlocked** The Princeton Review, 2016-11-29 UNLOCK THE SECRETS OF CHEMISTRY with THE PRINCETON REVIEW High School Chemistry Unlocked focuses on giving you a wide range of key lessons to help increase your understanding of chemistry With this book you ll move from foundational concepts to complicated real world applications building confidence as your skills improve End of chapter drills will help test your comprehension of each facet of chemistry from atoms to alpha radiation Don t feel locked out Everything You Need to Know About Chemistry Complex concepts explained in straightforward ways Walk throughs of sample problems for all topics Clear goals and self assessments to help you pinpoint areas for further review Guided examples of how to solve problems for common subjects Practice Your Way to Excellence 165 hands on practice questions seeded throughout the chapters and online Complete answer explanations to boost understanding Bonus online questions similar to those you ll find on the AP Chemistry Exam and the SAT Chemistry Subject Test High School Chemistry Unlocked covers Building blocks of matter Physical behavior of matter Chemical bonding Chemical reactions Stoichiometry Solutions Acids and bases Equilibrium Organic chemistry Radioactivity and more *Basic Chemistry Problems for High School* John



W. Sulcoski, Stephen Rituper, 1963      **Learning with Understanding in the Chemistry Classroom** Iztok Devetak, Saša Aleksij Glazar, 2014-01-14 This volume offers a critical examination of a variety of conceptual approaches to teaching and learning chemistry in the school classroom Presenting up to date research and theory and featuring contributions by respected academics on several continents it explores ways of making knowledge meaningful and relevant to students as well as strategies for effectively communicating the core concepts essential for developing a robust understanding of the subject Structured in three sections the contents deal first with teaching and learning chemistry discussing general issues and pedagogical strategies using macro sub micro and symbolic representations of chemical concepts Researchers also describe new and productive teaching strategies The second section examines specific approaches that foster learning with understanding focusing on techniques such as cooperative learning presentations laboratory activities multimedia simulations and role playing in forensic chemistry classes The final part of the book details learner centered active chemistry learning methods active computer aided learning and trainee chemistry teachers use of student centered learning during their pre service education Comprehensive and highly relevant this new publication makes a significant contribution to the continuing task of making chemistry classes engaging and effective      The Science Teacher , 1994      Must Know High School Chemistry Mary Millhollon, Richard H. Langley, 2019-06-28 A UNIQUE NEW APPROACH THAT S LIKE A LIGHTNING BOLT TO THE BRAIN You know that moment when you feel as though a lightning bolt has hit you because you finally get something That s how this book will make you react We hope Each chapter makes sure that what you really need to know is clear right off the bat and sees to it that you build on this knowledge Where other books ask you to memorize stuff we re going to show you the must know ideas that will guide you toward success in chemistry You will start each chapter learning what the must know ideas behind a chemistry subject are and these concepts will help you solve the chemistry problems that you find in your classwork and on exams Dive into this book and find 250 practice questions that mirror what you will find in your classwork and on exams A bonus app with flashcards that will reinforce what you ve learned Extensive examples that drive home essential concepts An easy access setup that allows you to jump in and out of subjects Chemistry topics aligned to national and state education standards Special help for more challenging chemistry subjects including the mole concept stoichiometry and solutions We re confident that the must know ideas in this book will have you up and solving chemistry problems in no time or at least in a reasonable amount of time      **Elementary Principles of Chemical Processes** Richard M. Felder, Ronald W. Rousseau, Lisa G. Bullard, 2020-08-11 This best selling text prepares students to formulate and solve material and energy balances in chemical process systems and lays the foundation for subsequent courses in chemical engineering The text provides a realistic informative and positive introduction to the practice of chemical engineering      Illinois Chemistry Teacher , 1992      **Improving Student Comprehension of Stoichiometric Concepts** Connie Lynn Bannick Kemner, 2007      **Science** John Michels (Journalist), 2010      APS PGT Exam PDF-AWES-Army Public School PGT



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*Chemistry* John W. Moore,Conrad L. Stanitski,Peter C. Jurs,2002 CHEMISTRY THE MOLECULAR SCIENCE is intended to help students develop a broad overview of chemistry and chemical reactions an understanding of the most important concepts and models that chemists and those in chemistry related fields use an appreciation of the many ways chemistry impacts our daily lives the ability to apply the facts concepts and models of chemistry appropriately to new situations in chemistry other sciences and engineering and to other disciplines

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In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Limiting Reactant Problems And Answers High School**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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