

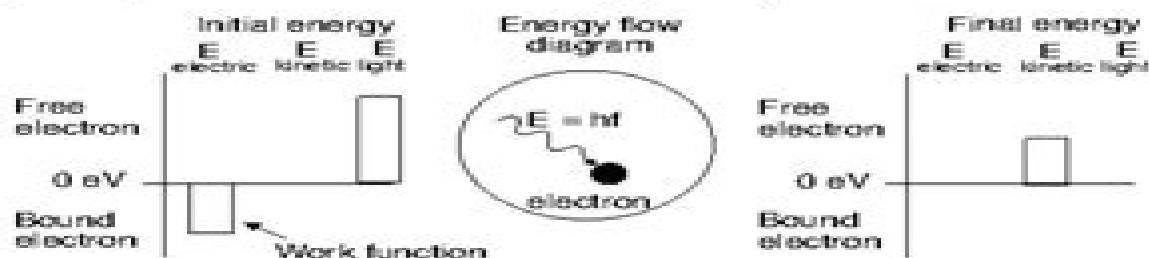
Worksheet 1b: Quantitative Photoelectric Effect

Planck's Constant, $h = 6.63 \times 10^{-34} \text{ J sec}$ or $4.14 \times 10^{-15} \text{ eV sec}$ speed of light, $c = 3 \times 10^8 \text{ m/s}$

electron-volt, $1 \text{ eV} = 1.60 \times 10^{-19} \text{ J}$

Example:

A metal with a work function of 4.5 eV is lit by incident light of frequency $3.5 \times 10^{15} \text{ Hz}$. Draw an energy diagram and predict the maximum kinetic energy of the ejected electrons.



Energy conservation for the photoelectric effect:

$$\text{work function} + \text{light energy} = \text{maximum electron kinetic energy}$$

$$E_{\text{photoelectric}} = \text{work function} = -4.5 \text{ eV}$$

The negative value means that energy must be transferred to the electron just to free it from the metal.

$$E_{\text{light}} = hf = (6.63 \times 10^{-34} \text{ J}\cdot\text{s})(3.5 \times 10^{15} \text{ Hz}) = 2.3 \times 10^{-18} \text{ J} \quad (1 \text{ eV} / 1.60 \times 10^{-19} \text{ J}) = 14.5 \text{ eV}$$

$$E_{\text{kinetic}} = 14.5 \text{ eV} - 4.5 \text{ eV} = 10.0 \text{ eV}$$

Problems:

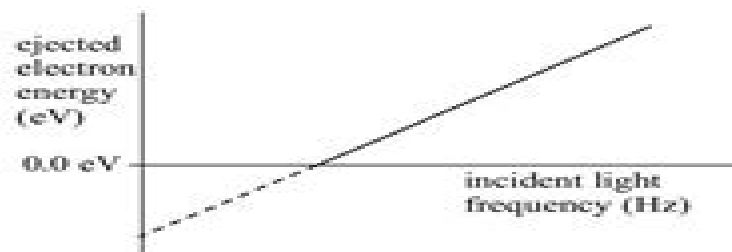
1. Refer to the graph as you answer the following questions.

a. What is the value and name of the slope of the line?

b. What is the name and meaning of the x-intercept?

c. What is the name and meaning of the y-intercept?

d. Describe the procedure used to collect the data for this graph.



Light Lab Answer Sheet

Laurie E. Westphal



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Sound and Light Michael J. Padilla, Ioannis Miaoulis, Martha Cyr, Jay M. Pasachoff, 2002

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Table of Contents Light Lab Answer Sheet

1. Understanding the eBook Light Lab Answer Sheet
 - The Rise of Digital Reading Light Lab Answer Sheet
 - Advantages of eBooks Over Traditional Books
2. Identifying Light Lab Answer Sheet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Light Lab Answer Sheet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Light Lab Answer Sheet
 - Personalized Recommendations
 - Light Lab Answer Sheet User Reviews and Ratings
 - Light Lab Answer Sheet and Bestseller Lists
5. Accessing Light Lab Answer Sheet Free and Paid eBooks

- Light Lab Answer Sheet Public Domain eBooks
- Light Lab Answer Sheet eBook Subscription Services
- Light Lab Answer Sheet Budget-Friendly Options
- 6. Navigating Light Lab Answer Sheet eBook Formats
 - ePub, PDF, MOBI, and More
 - Light Lab Answer Sheet Compatibility with Devices
 - Light Lab Answer Sheet Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Light Lab Answer Sheet
 - Highlighting and Note-Taking Light Lab Answer Sheet
 - Interactive Elements Light Lab Answer Sheet
- 8. Staying Engaged with Light Lab Answer Sheet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Light Lab Answer Sheet
- 9. Balancing eBooks and Physical Books Light Lab Answer Sheet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Light Lab Answer Sheet
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Light Lab Answer Sheet
 - Setting Reading Goals Light Lab Answer Sheet
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Light Lab Answer Sheet
 - Fact-Checking eBook Content of Light Lab Answer Sheet
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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