

Chem 101 — General Chemistry
Practice Final Exam

Notes:

$k_1 = 6.426 \times 10^{-12} \text{ s}^{-1}$ (forward) & k_{-1} (reverse)

$k_2 = 5.00 \times 10^7 \text{ s}^{-1}$ (reverse of k_1)

$k_3 = 1.000 \times 10^{-11} \text{ s}^{-1}$ (forward) & k_{-3} (reverse)

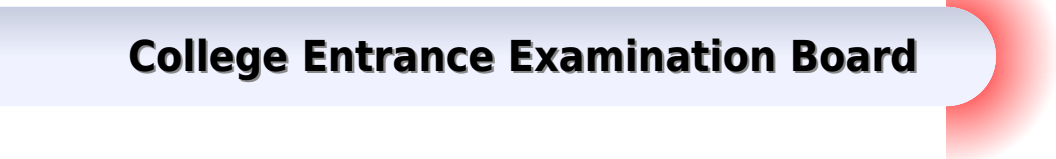
Strategies of Solvers (8 points each)

Identify the letter of the choice that best completes the statement or answers the question and legibly write the letter in the space preceding the problem number.

- _____ 1. Which of the following is not a colligative?
- a. mol ratio
 - b. molal ratio
 - c. mass
 - d. ΔT
 - e. molal ratio
- _____ 2. Which of the following statements concerning the photo effect?
- a. Chemical reactions involve only the electrons.
 - b. The above correct position, neutrons and electrons.
 - c. Photons are found in waves along the surface.
 - d. The surface contains both protons and neutrons.
 - e. Most of the mass of the atom is found in the nucleus.
- _____ 3. Which of the following statements about two isotopes is false?
- a. They will have the same charge on the nucleus.
 - b. They will have different numbers of neutrons.
 - c. They will have essentially the same chemical reactivity.
 - d. They will have the same atomic number.
 - e. They will have the same atomic weights.
- _____ 4. Which element can be classified as an alkaline earth metal?
- a. Al
 - b. Mg
 - c. Ag
 - d. Na
 - e. Fe
- _____ 5. Which combination below corresponds to a cation with a +1 charge?
- a. 7 protons and 8 electrons
 - b. 11 protons and 10 electrons
 - c. 15 protons and 14 electrons
 - d. 17 protons and 16 electrons
 - e. 19 protons and 18 electrons

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